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$$\begin{array}{l} 1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3 \\ 2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2 \\ 2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} \\ \text{Cl}_2 + 2 \text{ H}_2 \rightarrow 2 \text{ HCl} \\ 2 \text{ H}_2\text{O} + 5 \text{ Pb} \rightarrow 2 \text{ H}_2 + \text{O}_2 + 5 \text{ PbO} \\ 2 \text{ HCl} + 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 \rightarrow 2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2\text{O} + 2 \text{ CH}_4 \\ 2 \text{ O}_2 + 1 \text{ CO}_2 + 2 \text{ H}_2\text{O} + 8 \text{ C} + 3 \text{ H}_8 + 5 \text{ O}_2 + 3 \text{ CO}_2 + 4 \end{array}$$

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$$\begin{array}{l} 1 \text{ O}_2 + 2 \text{ NH}_3 \rightarrow 3 \text{ H}_2\text{O} + 2 \text{ N}_2 + 3 \text{ H}_2 \\ 2 \text{ O}_2 + 2 \text{ H}_2\text{O} + 2 \text{ RbH} + 2 \text{ RbOH} + 2 \text{ H}_2 + 5 \text{ Cl}_2 + 2 \text{ KBr} + \text{KCl} + \text{Br}_2 + 6 \text{ N}_2 + 2 \text{ H}_2 + \text{NH}_3 + 7 \text{ N}_2\text{F}_6 + 3 \text{ Ca} + \text{N}_2\text{O} + \text{C}_2\text{H}_5\text{OH} \end{array}$$

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$$\begin{array}{l} 38 \text{ Fe} + 2 \text{ O}_3 + 3 \text{ CO} + 2 \text{ Fe}_3\text{CO}_2 + 39 \text{ H}_2 + 3 \text{ PO}_4 + 3 \text{ BaO} + 3 \text{ PO}_4 + 2 \text{ H}_2 + 40 \text{ Mg} + \text{NH}_4 + 4 \text{ PO}_4 + 2 \text{ P}_2 + \text{O}_7 + 2 \text{ NH}_3 + 3 \text{ H}_2 + \text{O}_4 + 1 \text{ Al}_2 + 3 \text{ H}_2 + \text{SO}_4 + \text{Al}_2 + \text{SO}_4 + 3 \text{ H}_2 + 2 \text{ Al}_2 + 2 \text{ P}_4 + \text{O}_{10} + 3 \text{ HCl} + \text{POCl}_3 + 3 \text{ HPO}_3 + 4 \text{ HNO}_2 + 2 \text{ O}_3 + 2 \text{ HNO}_2 \end{array}$$

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