



MY FIRST FIVE YEARS™

Pre-schooler



The My First Five Years approach

It's all about play!

If you haven't heard of us before, we at My First Five Years like to shout loudly about the importance of play! Our first belief is that play is not in opposition to work. Play can truly can be utterly joyful and fun, but play can also be a serious thing even for a young child. It can be scientific, mathematical, creative, and involves testing ideas, making connections, adopting approaches, persevering and much more.

The prevalent concerns about 'failing education' and performing goals within certain timetables can lead to parents feeling they need to adopt more formal and adult-driven learning approaches which tend to hurry children, even pre-school age, through structured experiences, squeezing time for play - especially free play - out of their lives.

We want to reset the balance.

We know that play can be so many things and its importance cannot be underestimated. Lots of words can be used to describe play: mindless, challenging, absorbing, exploratory, daring, evolving, quiet, sometimes social, creative, imaginative, skillful, emotional... we could go on forever!

Watching your child will show you that their drive to play is natural and irresistible, and hidden to the rest of us in the most unlikely, and least obviously playful, situations. At this time your child's curiosity and need to actively explore will be high, and they will have the gravity/magnetism to pull their explorations and experience of the world, as they begin to forge relationships and build new community outside of the home.

Recognising that your child loves being outside, many of our activities support this. Playing outdoors is not only essential for your child to understand, value, enjoy and protect our natural world - it also gives them a treasure trove of unique experiences and play opportunities.



This book, and often this period of childhood, is called 'pre-school' but the things your child is learning now will provide them with the skills that go far beyond just preparing for school. One of our favourite quotes of My First Five Years is that 'Children are human beings', not human 'becomings' and that is especially true now as outside pressures to get ready for the next step become stronger.



By focusing on living in the now, understanding how your child is developing and noticing what helps them most at this moment, they will be building on their solid foundation of skills which will help them to thrive throughout their lives.

All the activities in this book are designed for children who are growing out of the toddler years and who would benefit from activities that might sustain their attention for a little longer whilst supporting them in developing the social skills, problem-solving and reasoning that will be useful as they grow up and become more independent.

We don't want to hurry you or your child to be 'ready' for the next stage, rather we argue for an unhurried relationship with time to make the most of deep development opportunities and the daily flow of life. So, slow into the activities and embrace doing them in the moment, alongside your child.



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My First Five Years My First Five Years

Ying Liu



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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 °C and 64 °C respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1-2 mm of the capillary tube end c

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affected by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 °C

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the melting point apparatus compound a had a melting point range of 114.6-115.5 °C thus revealing it to be acetanilide compound b had a melting point range of 122.6-124.5 °C thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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