

The Reader's Digest

# *Merry Christmas* Songbook



# Merry Christmas Sound Book Reader Digest William L Simon

**Yi-Tong Ma**



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web the rate law for a particular reaction is rate  $k \times x^2$  in an experiment the initial rate of the reaction is determined to be 0.080 mol l s when the initial concentration of x is

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web a rate law shows how the rate of a chemical reaction depends on reactant concentration for a reaction such as  $aA \rightarrow bB$  products the rate law generally has the form  $\text{rate} = k[A]^n$

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web the reaction orders in a rate law describe the mathematical dependence of the rate on reactant concentrations referring to the generic rate law above the reaction is  $m$

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web each reactant in the rate limiting step of a reaction is assigned an order typically zeroth 1st or 2nd the order is the sum of all individual orders the rate of a reaction is

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web the rate law or differential rate law relates the rate of a reaction to the concentration or pressure of the reactants the rate of a reaction is proportional to the concentration or

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web oct 27 2022 the rate law for the reaction  $\text{ce h}_2 \text{ g ce } 2\text{no g ce n}_2\text{o g ce h}_2\text{o g}$  nonumber has been experimentally determined to be rate

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web 1 use table 1 to determine the reaction order for hcl 2 use table 2 to determine the reaction order for  $\text{na}_2\text{s}_2\text{o}_3$  remember you want to see what happens to the reaction

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