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web mass 32 g mass 16 g mass 3 2 g 2 for jars d e f use the number of molecules in the circle to determine the number of moles of  $O_2$  in the jar then calculate the mass of the gas in the jar d e f 12 mol  $O_2$  2 mol  $O_2$  6 mol  $O_2$  mass 384 g mass 64 g mass 192 g 3 jars g h and i contain ammonia gas  $NH_3$

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web jul 4 2019 solution from the balanced equation it is known that 1 mol  $N_2$  2 mol  $NH_3$  use the periodic table to look at the atomic weights of the elements and calculate the weights of the reactants and products 1 mol of  $N_2$  28 0 g 2 mol of  $NH_3$  is 34 0 g 3 1 0 g 17 0 g

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web solution average atomic mass of Li 6 941 amu  $\times$  6 0151 amu  $\times$  7 0160 amu 6 941 1 0009  $\times$  7 0160 1 0009  $\times$  0 075  $\times$  0 075  $\times$  0 075 corresponds to a natural abundance of 6 Li of 7 5 percent the natural abundance of  $^7Li$  is  $\times$  0 925 or 92 5 percent 3 7 the unit factor required is 6 022

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web a molar mass  $\text{Li}_2\text{CO}_3$  2 6 941 g 12 01 g 3 16 00 g 73 89 g b molar mass  $\text{CS}_2$  12 01 g 2 32 07 g 76 15 g c molar mass  $\text{CHCl}_3$  12 01 g 1 008 g 3 35 45 g 119 37 g d molar mass  $\text{C}_6\text{H}_8\text{O}_6$  6 12 01 g 8 1 008 g 6 16 00 g 176 12 g e molar mass  $\text{KNO}_3$  39 10 g 14 01 g 3 16 00 g 101 11 g

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web it is a small step from mole mass calculations to mass mass calculations if we start with a known mass of one substance in a chemical reaction instead of a known number of moles we can calculate the corresponding masses of other substances in the reaction

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web jul 4 2022 to calculate the mass of titanium metal that you can obtain multiply the number of moles of titanium by the molar mass of titanium 47 867 g mol moles ti mass ti times molar mass ti 4 12 cancel mol ti times frac 47 867 g ti 1 mol ti 197 g ti notag

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web 41 42 chemistry ch 3 mass relationships in chemical reactions molecular mass of  $\text{NO}_2$  atomic mass of n 2 atomic mass of o 14 01 ainu 2 16 00 ainu 46 01 amu atomic mass of  $\text{N}_2\text{O}_5$  2 atomic mass of n 5 atomic mass of o 2 14 01 ainu 5 16 00 ainu 108 02 amu example 3 2 exercise 3 3