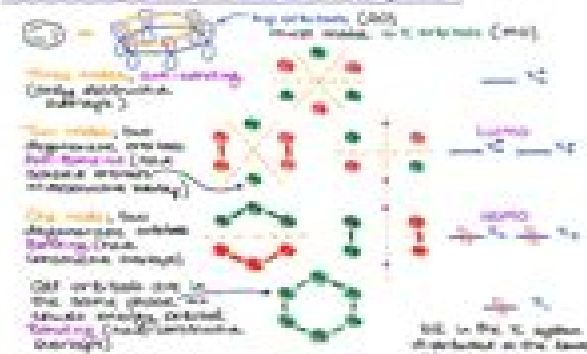


# Mechanism of Reaction in Aromatic Compounds



**Final Cycle Diagram** - shows the overall mechanism of electrophilic aromatic substitution.

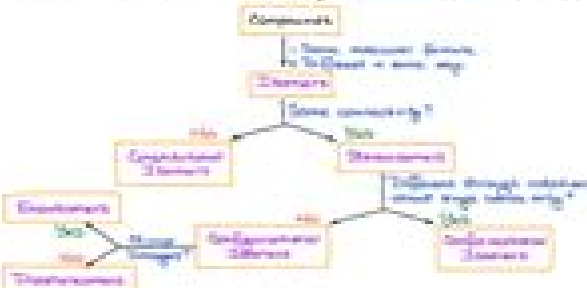
A cyclic compound is considered to be a cyclic, with a great deal of resonance.



Due to the resonance, the average bond length in benzene is 1.39 Å, which is intermediate between a single and a double bond.

## Summary of Stereochemical Relationships

- Constitutional Isomers** - completely different molecules
- Conformational Isomers** - same molecule that only differ in 3D shape due to free rotation around single bonds
- Enantiomers** - non-superimposable mirror images that share physical & chemical properties in achiral environment
- Diastereomers** - non-superimposable, non-mirror images that have different physical & chemical properties
- Meso Compounds** - non-chiral due to the plane of symmetry, however, have chiral centers



## Stereochemical Relationships of Enantiomers

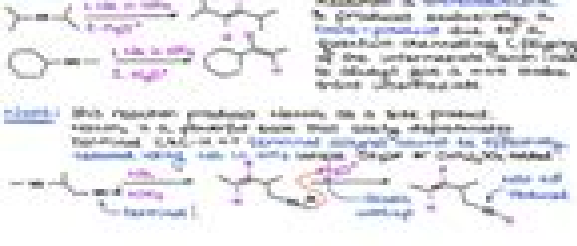
Enantiomers are mirror images, non-superimposable, and have identical physical properties in achiral environment, they are the only pair of molecules that are mirror images of each other.

Enantiomers have identical physical & chemical properties in achiral environment, they are the only pair of molecules that are mirror images of each other.

## Reaction of Alkylated with Nitration in Liquid Phase

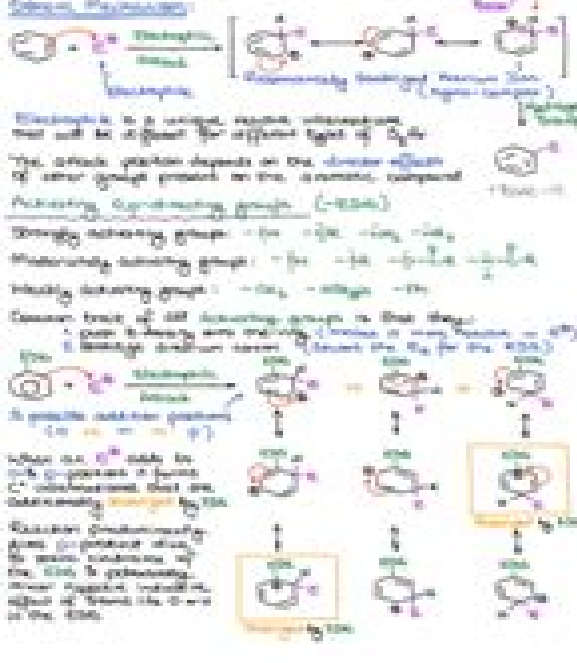


**Final Cycle Diagram** - shows the overall mechanism of electrophilic aromatic substitution.

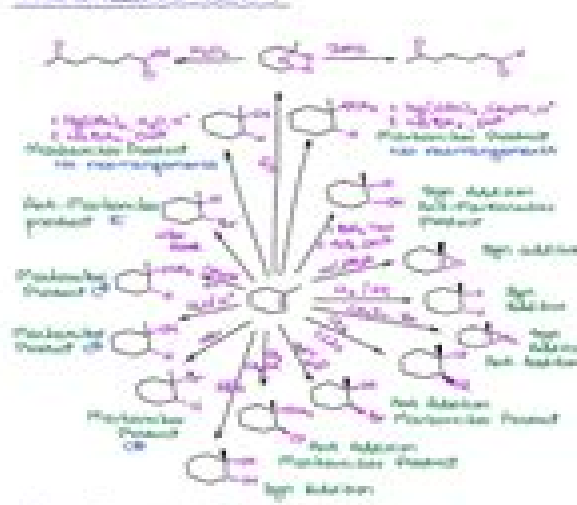


Due to the resonance, the average bond length in benzene is 1.39 Å, which is intermediate between a single and a double bond.

## Electrophilic Aromatic Substitution Reactions (EAS)



## Alkylated Aromatic Substitution

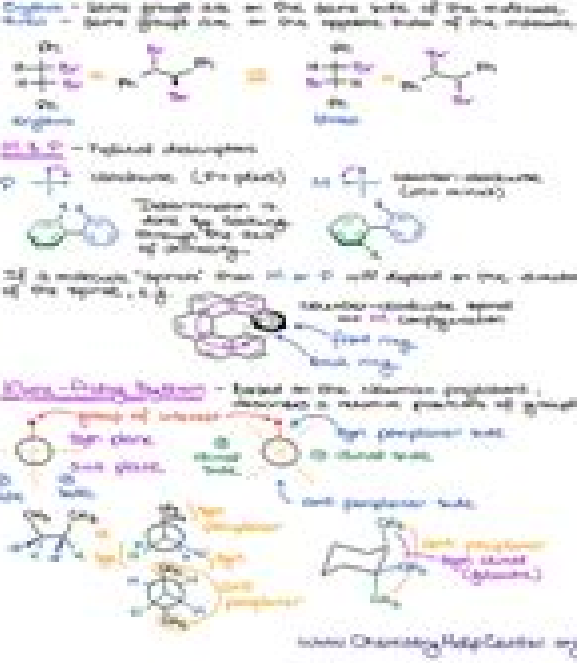


**Final Cycle Diagram** - shows the overall mechanism of electrophilic aromatic substitution.



Due to the resonance, the average bond length in benzene is 1.39 Å, which is intermediate between a single and a double bond.

## Reaction of Alkylated with Nitration in Liquid Phase



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