



| Atomic Number | Element | Isotope | Abundance |
|---------------|---------|---------|-----------|
| 1             | H       | 1       | 99.99     |
| 5             | B       | 10      | 19.9      |
| 6             | C       | 11      | 80.1      |
| 7             | N       | 12      | 98.9      |
| 8             | O       | 13      | 1.1       |
| 9             | F       | 14      | 99.6      |
| 15            | P       | 16      | 99.8      |
| 16            | S       | 17      | 100       |
| 17            | Cl      | 18      | 100       |
| 35            | Br      | 32      | 94.99     |
| 53            | I       | 33      | 0.8       |
|               |         | 34      | 4.2       |
|               |         | 35      | 75.8      |
|               |         | 37      | 24.2      |
|               |         | 79      | 50.7      |
|               |         | 81      | 49.3      |
|               |         | 127     | 100       |

| MW    | Species                          |
|-------|----------------------------------|
| 17    | $\text{NH}_3$                    |
| 18    | $\text{H}_2\text{O}$             |
| 20    | $\text{HF}$                      |
| 28    | $\text{CH}_2\text{CH}_2$         |
| 32    | $\text{CH}_3\text{OH}$           |
| 34    | $\text{H}_2\text{S}$             |
| 36/38 | $\text{HCl}$                     |
| 46    | $\text{EtOH}$                    |
| 62    | $\text{Me}_2\text{CD}_2\text{H}$ |
| 80/82 | $\text{HBr}$                     |

MS isotopes & common pattern pieces

# Organic Chemistry Ir And Nmr Cheat Sheet

**Yijin Wang**



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