



2. In solvent like  $H_2O$



**Vicinal dibromide formation :**

**Halohydrin formation**

- It is a **regioselective** reaction
- The nucleophilic attack of the water occurs at the more substituted carbon atom.

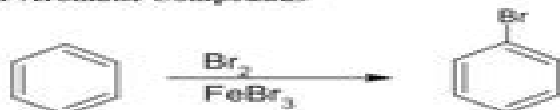
## 2. With Alkynes

- It is an **anti-addition** reaction:
- 1. In solvent like  $CCl_4$ ,  $CH_2Cl_2$  etc



**Tetrabromide formation**

## 3. With Aromatic Compounds

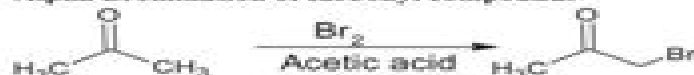


## 4. Photo halogenation of alkanes



**Major**

## 5. Alpha Bromination of carbonyl compounds



## 6. Oxidation of Monosaccharides

1. Bromine water is a mild oxidizing agent
2. selectively oxidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.



## 7. Haloform Reaction

1. converts methyl ketones to Haloform ( $CHX_3$  ( $X=Cl, Br, I$ )) and salt of carboxylic acids.



## 8. The Hoffmann Bromamide degradation

1. It converts primary amide to a primary amine with one less carbon atom.



## 9. Hell-Volhard-Zelinski (or HVZ) reaction



## 5. $CN^-$

### Cyanide Ion and Nitrile Group

#### 1. Conversion of Nitriles to a Primary Amine



#### 2. Conversion of Nitriles to Aldehydes



#### 3. Reaction of Nitriles with Grignard reagent



#### 4. Hydrolysis of Nitriles to Carboxylic Acids



# Organic Chemistry Laboratory Study Guide

**Charles Richmond Henderson**



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