

Chapter 5 The Periodic Table

Section 5.1 Organizing the Elements

(pages 126–129)

This section explains how *Mendeleev* organized elements into a periodic table. It also discusses the predictions he made about undiscovered elements and later the discovery of those elements supported his version of the table of the table.

Reading Strategy (page 126)

Identifying Main Ideas As you read, complete the table by identifying the main idea for each topic. For more information on this reading strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Topic	Main Idea
Mendeleev's proposal	Mendeleev arranged the elements in order of increasing mass so that elements with similar properties were in the same column.
Mendeleev's prediction	Mendeleev used the properties of existing elements to predict properties of undiscovered elements.
Evidence supporting Mendeleev's table	The close match between Mendeleev's predictions and the actual properties of new elements showed that his periodic table could be.

The Search for Order (page 126)

1. Is the following sentence true or false? The first elements to be identified were mainly gases. false
2. As the number of known elements grew, so did the need to organize them into groups based on their properties.
3. Circle the letter of each category that the French chemist Antoine Lavoisier used to classify elements.

☒ a. gases
c. liquids

☒ b. metals
☒ d. nonmetals

Mendeleev's Periodic Table (pages 127–128)

4. Is the following sentence true or false? Mendeleev needed to organize information about 63 elements. is so
5. Mendeleev's strategy for classifying elements was modeled on a(n) card game or solitaire.
6. Circle the letter of each type of information Mendeleev knew about each element.

☒ a. name

☐ b. number of protons

☒ c. relative mass

☒ d. properties

Organizing The Elements Workbook

Carl Louis Stotz



Organizing The Elements Workbook:

The Elements Book DK, 2017-04-04 Richly illustrated with over a thousand photos and dazzling details of the elements that make up the physical world Written in association with the renowned Smithsonian Institution Does your little chemist have questions about the stuff that everything is made of This visual reference book covers each of the 118 elements and includes a glossy pull out poster of the periodic table This encyclopedia is a superb introduction to the subject of chemistry Written with kids ages 9 to 12 in mind using easy to understand language and straightforward fun facts There s information on the scientists that made the first discoveries and spectacular photos of large natural features along with a simple explanation of what an atom is Find out which of the things we see every day contain these common and unusual elements There s so much to discover about different elements Explore their atomic structure with the number of electrons protons and neutrons and the three states of solid liquid or gas Kids will learn that the copper used in computer motherboards is also what the Statue of Liberty is made of and why it s green Also learn about elements like zinc why Japan s Akashi Bridge is coated in zinc and why zinc is used in the soles of boots to make the rubber tougher Each element is shown in its pure form in a stunning series of photos that will keep children engrossed in elemental science The poster included with this education book is an added learning tool that shows how the elements are arranged on the periodic table It s easier than ever to look up the basics of chemistry From Ac to Xe and all the elements between The multitude of photos in this appealing format makes learning the fundamentals of chemistry simple and enjoyable This visual reference guide provides the reader with an overview of the most fascinating facts about the elements within us and around us Concise and bite sized information makes it easy for young scholars to follow Eye catching and captivating photos of raw elements and what they are used in

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