



Double click to open the Property Sheet

Navigation Pane

Switchboard

Form Header

 **Welcome to The Sapphire Sales Application**

Detail

Click on the icons below to access data entry forms and reports

 **Open Sales Entry Form**

 **Open Sales Area Form**

 **Open Individual Area Report**

 **Open Combined Area Report**

 **View Sales Record Form**

 **View Sales Record Form**

Form Footer

...open the form in Design View. Open the Property Sheet and select the Format tab. Change "Navigation Buttons" to "No."

Property Sheet

Selection type: Form

Form

Format Data Event Other All

Caption	Single Form
Default View	Single Form
Allow Form View	Yes
Allow Datasheet View	Yes
Allow Print/able View	Yes
Allow Print/Chart View	Yes
Allow Layout View	Yes
Picture	(Image)
Picture Tiling	No
Picture Alignment	Top Left
Picture Type	Embedded
Picture Size Mode	Stretch Horizontal
Width	6.7708"
Auto Center	No
Auto Resize	Yes
Fit to Screen	Yes
Border Style	Scalable
Record Selectors	Yes
Navigation Buttons	No
Navigation Caption	
Dividing Lines	No
Scroll Bars	Both
Control Box	Yes
Close Button	No
Min Max Buttons	Both Enabled
Movable	No
Split Form Size	Auto
Split Form Orientation	Datasheet on Top
Split Form Splitter Bar	Yes
Split Form Datasheet	Allow Edits
Split Form Printing	Form Only
Save Splitter Bar Position	Yes
Subdatasheet Expanded	No
Subdatasheet Height	0"
Grid X	24
Grid Y	24
Layout For Print	No
Orientation	Left-to-Right
Palette Source	(Default)

Microsoft Access 2007 Custom Guide

John Pierce



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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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 $\text{AgNO}_3 \text{ aq} + \text{KCl aq} \rightarrow \text{AgCl s} + \text{KNO}_3 \text{ aq}$ total ionic $\text{Ag}^+ \text{ aq} + \text{NO}_3^- \text{ aq} + \text{K}^+ \text{ aq} + \text{Cl}^- \text{ aq} \rightarrow \text{AgCl s} + \text{K}^+ \text{ aq} + \text{NO}_3^- \text{ aq}$ net ionic $\text{Ag}^+ \text{ aq} + \text{Cl}^- \text{ aq} \rightarrow \text{AgCl s}$

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water $\text{H}^+ \text{ aq} + \text{OH}^- \text{ aq} \rightarrow \text{H}_2\text{O l}$ the spectator ions in KOH aq and $\text{HNO}_3 \text{ aq}$ are $\text{K}^+ \text{ aq}$ and $\text{NO}_3^- \text{ aq}$ how to identify and list the

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