

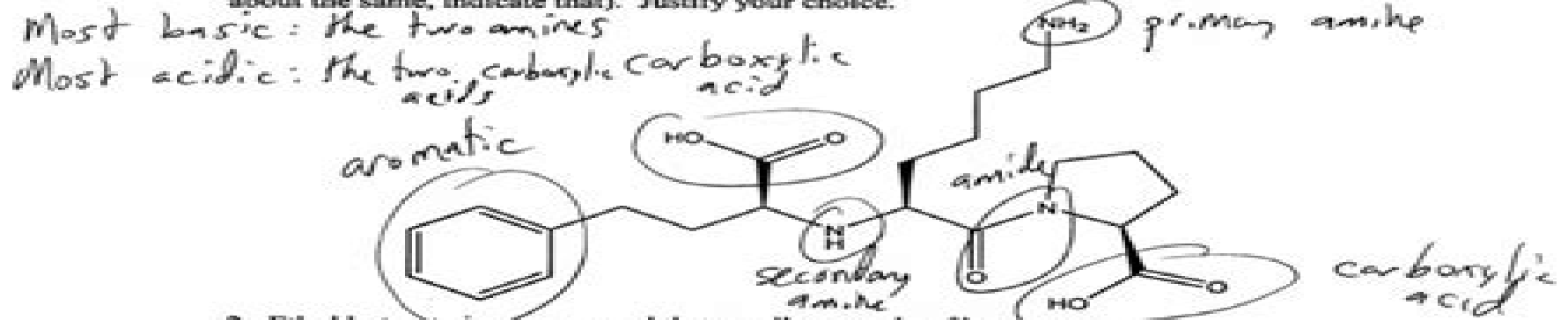
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## Exam 1

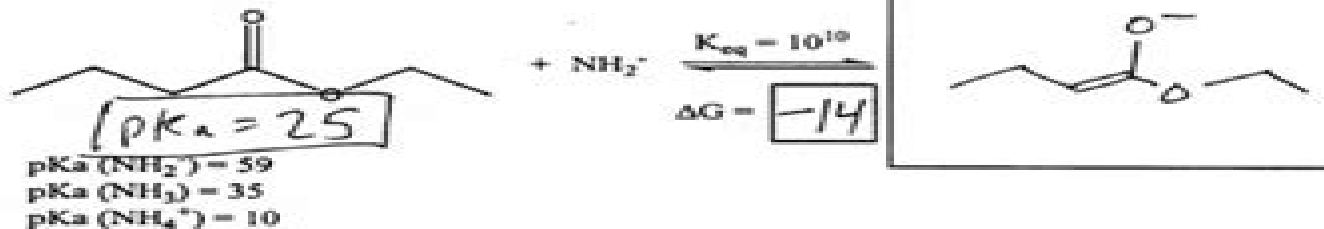
Tuesday, October 27, 2009

Answer as many questions as you can in the time allowed. Please be as clear and concise as possible.

1. Zestril is a bestselling medication used to treat high blood pressure.
  - a. Identify all the functional groups in morphine. Classify as primary, secondary, tertiary where relevant.
  - b. Identify the most acidic site and the most basic site in Zestril (if more than one will be about the same, indicate that). Justify your choice.



2. Ethyl butyrate is a compound that smells strongly of banana.
  - a. Ethyl butyrate is somewhat acidic. Draw its conjugate base.
  - b. Ethyl butyrate can be deprotonated with  $\text{NH}_2^-$ . The  $K_{\text{eq}}$  for this reaction is  $10^{10}$ . What is the  $\text{pK}_a$  of ethyl butyrate? What is the  $\Delta G$  for this reaction?



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**RC Schank**



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## **Table of Contents Organic Chem 1 Exam Une**

1. Understanding the eBook Organic Chem 1 Exam Une
  - The Rise of Digital Reading Organic Chem 1 Exam Une
  - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chem 1 Exam Une
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Organic Chem 1 Exam Une
  - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chem 1 Exam Une
  - Personalized Recommendations
  - Organic Chem 1 Exam Une User Reviews and Ratings
  - Organic Chem 1 Exam Une and Bestseller Lists
5. Accessing Organic Chem 1 Exam Une Free and Paid eBooks
  - Organic Chem 1 Exam Une Public Domain eBooks
  - Organic Chem 1 Exam Une eBook Subscription Services
  - Organic Chem 1 Exam Une Budget-Friendly Options

6. Navigating Organic Chem 1 Exam Une eBook Formats
  - ePub, PDF, MOBI, and More
  - Organic Chem 1 Exam Une Compatibility with Devices
  - Organic Chem 1 Exam Une Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Organic Chem 1 Exam Une
  - Highlighting and Note-Taking Organic Chem 1 Exam Une
  - Interactive Elements Organic Chem 1 Exam Une
8. Staying Engaged with Organic Chem 1 Exam Une
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Organic Chem 1 Exam Une
9. Balancing eBooks and Physical Books Organic Chem 1 Exam Une
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Organic Chem 1 Exam Une
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Organic Chem 1 Exam Une
  - Setting Reading Goals Organic Chem 1 Exam Une
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Organic Chem 1 Exam Une
  - Fact-Checking eBook Content of Organic Chem 1 Exam Une
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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