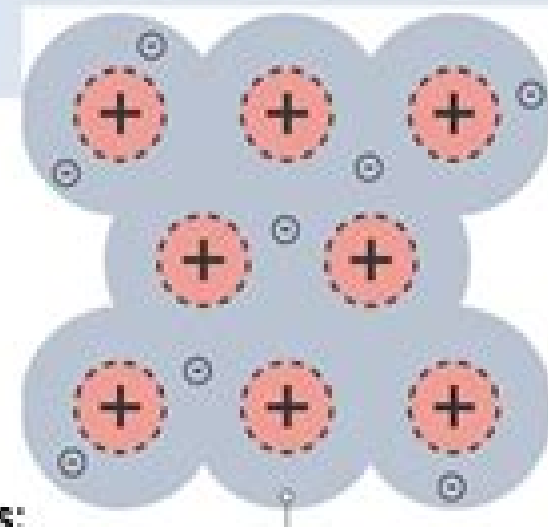


What is metallic bonding?

Questions:

1. What charge do electrons have?
2. What charge do metal ions have?
3. Copy and label the diagram of a metallic bond with the name of the particles.
4. What force holds the metal cations and electrons together?
5. Describe the structure of metals in terms of the particles present in their structures.
6. Why do metals have high melting points?



Answers:

1. Negative
2. Positive
3. Small circle – negative electron;
Large circle – positive metal cations
4. Electrostatic
5. A large lattice containing positive ions/cations surrounded by electrons
6. Strong electrostatic forces between cations and electrons which requires a lot of energy to break

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LL Leslie



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