

## CHEMISTRY I- Semester

Paper-I(T)    Course No.: CHEM-111    Marks: [55+15]    Lecture: 35 Hours

### *Organic Chemistry-I*

#### UNIT I

##### Structure and Bonding

Nature of Covalent bond and orbital representation of simple molecules (as listed below); hybridization of orbitals, implications of hybridizations on the concept of bond-length, bond-energy, bond-angles and shape of the molecules with the following examples:- VSEPR Theory.

(i)  $\text{CH}_4$ ,  $\text{H}_3\text{O}^+$ ,  $\text{CH}_3^-$ ,  $\text{RNH}_2$  (ii)  $\text{C}_2\text{H}_4$ ,  $\text{CH}_3^+$ ,  $\text{BF}_3$ ,  $\text{AlCl}_3$ , Carbonyl Compounds and (iii) Acetylene, R-CN, allene and ketone.

Electronegativity, polarity of bonds, inductive-effect in substituted aliphatic carboxylic acids, hydrogen-bonding and its effect on boiling point and solubility of organic molecules, intermolecular and intramolecular hydrogen bonding.

#### UNIT II

##### Electron-displacement effects in organic molecules and the basic concept of reaction mechanism

Role of inductive effect, Electromeric-effect, Mesomeric-effect or Resonance, hyperconjugation, aromaticity, hydrogen-bonding, curved arrow notation, drawing electron movements with arrows, half headed and double-headed arrows, homolytic and heterolytic bond-breaking, Types of reagents-electrophiles & nucleophiles. Energy considerations, reactive intermediates-carbocations, free-radicals, carbanions, carbenes, arynes, the stability of reaction intermediates.

Introduction to Aromaticity.

##### UNIT III Stereochemistry of organic molecules: Concept of Isomerism, types of Isomerism.

(a) Concept of homo and heterotopocity, prostereoisomerism.

(b) Geometrical Isomerism: E & Z system of nomenclature, geometrical isomerism in oximes and alicyclic compounds.

(c) Optical Isomerism : Elements of symmetry, molecular chirality, enantiomers, stereogeniccentre, optical activity, properties of enantiomers, chiral and achiral molecules with two stereogeniccentres, diastereo isomers, meso compounds, resolution of enantiomers, inversion and retention of configurations, Racemization. Relative and absolute configuration; chemoselectivity, regioselectivity, stereoselectivity(basic concept).

(d) Sequence rules, D & L and R & S system of nomenclature.

# Mcgraw Hill Organic Chemistry Solutions Manual Chandra

**Ying-Ying Zheng**



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