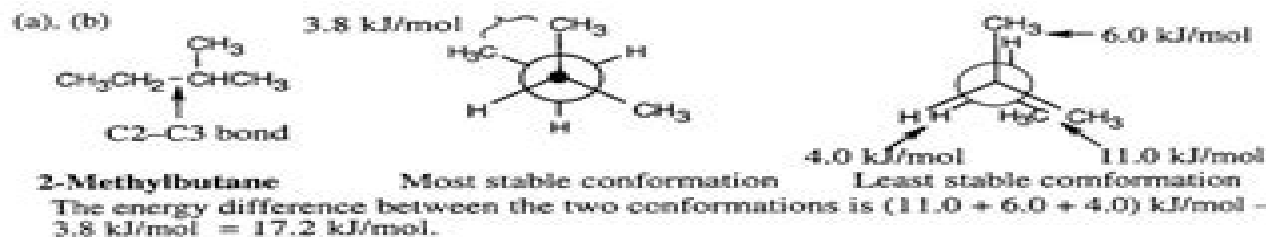
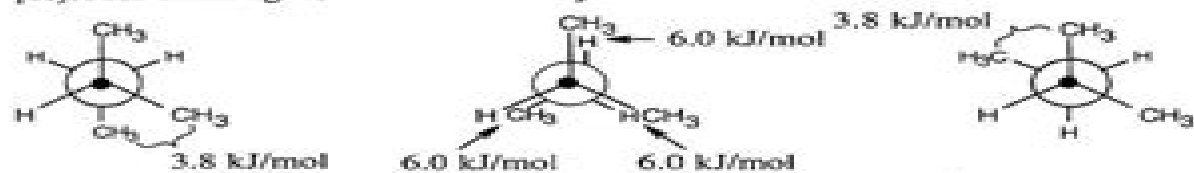


## 3.42



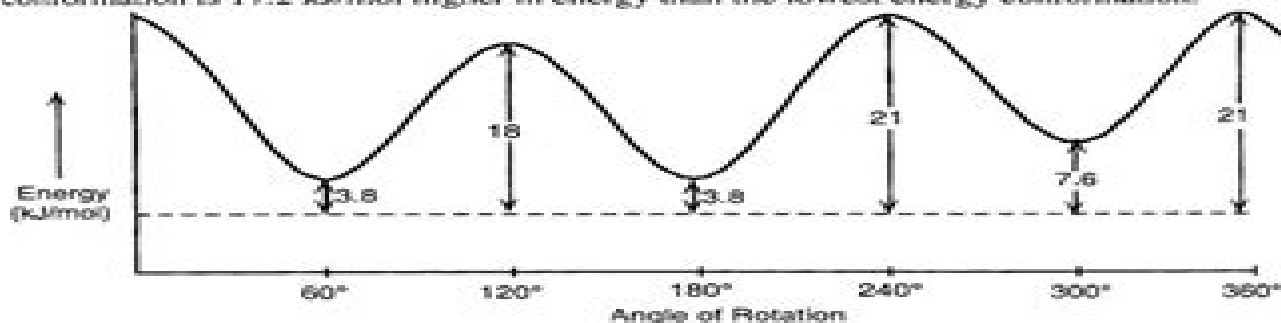
(c) Consider the least stable conformation to be at zero degrees. Keeping the front of the projection unchanged, rotate the back by  $60^\circ$  to obtain each conformation.



at  $60^\circ$ : energy = 3.8 kJ/mol at  $120^\circ$ : energy = 18.0 kJ/mol at  $180^\circ$ : energy = 3.8 kJ/mol



Use the lowest energy conformation as the energy minimum. The highest energy conformation is 17.2 kJ/mol higher in energy than the lowest energy conformation.



# **Mcmurry Organic Chemistry Solutions Manual Scribd**

**Francis Carey, Neil Allison**



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