



SPOT THE PERCENTAGE 1A ANSWERS

- 1) In a class, 65% of the children are girls. What percentage are boys?

Answer: 35%

- 2) In a packet of skittles, 27% are red. What percentage are not red?

Answer: 73%

- 3) In a group of children, 78% can swim. What percentage are non-swimmers?

Answer: 22%

- 4) Frazer scores 43% in a spelling test. What percentage did he get wrong?

Answer: 57%

- 5) Toronto FC have won 45% of their games and lost 42% of their games. What percentage have they tied?

Answer: 13%

- 6) Children had to pick their favourite sport from basketball, football and hockey. 35% chose basketball and 28% chose hockey. What percentage chose football?

Answer: 37%

- 7) 38% of the animals in a dog and cat rescue centre are cats. What percentage are dogs?

Answer: 62%

- 8) In a math test, 35% of the children scored an average score. What percentage scored above average?

Impossible question

- 9) In a study of students under 25 years old, 12% have not yet learned to drive. What percentage can drive?

Answer: 88%

- 10) In a survey 6% of adults did not know how to ride a bike. What percentage could ride?

Answer: 94%



Metric Conversion Word Problems Answers

AN Whitehead



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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web thus for reaction 9 2 in which 1 mole of h₃o and 1 mole of oh combine to form 2 moles of h₂o h 55 8 kj as shown note that the enthalpy change h hproducts hreactants is positive if heat is absorbed that is if hproducts

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