VANCOUVER COMMUNITY COLLEGE

Naming Inorganic Compounds 2

This worksheet contains numerous questions related to the naming of inorganic compounds. If you have not already done so, it is probably worth looking over the information given in the handout "Naming Inorganic Compounds".

EXERCISES

- A. Write the symbols for the following cations (positive ions):
 - 1) sodium
 - 2) calcium
 - copper (II)
 - 4) aluminum
 - 5) iron (III)
 - 6) zinc

- 7) potassium
 - 8) silver
 - 9) magnesium
 - 10) hydrogen
 - 11) ammonium
 - 12) manganese (IV)
- B. Write the symbols for the following anions (negative ions):
 - 1) chloride
 - 2) bromide
 - 3) iodide
 - 4) sulphide
 - 5) oxide
 - 6) sulphate
 - 7) sulphite

- 8) nitrate
- 9) nitrite
- 10) hydroxide
- 11) phosphate
- 12) carbonate
- 13) chlorate
- 14) chlorite
- C. Name the following binary molecular compounds composed of nonmetals:
 - 1) CO
 - 2) CO₂
 - 3) PBrs
 - 4) SO₂
 - 5) SO₃

- 6) C(2O7
- 7) H₂S_(g)
- 8) NF₃
- 9) N₂O₅
- 10) NO₂

Naming Inorganic Compounds Worksheet

Patrick Vollmar

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