

Tips for naming molecules

There are three steps, though that I have found useful over the years. The first concerns organic molecules in their simplest form, those for primary school years.

Remember that the name of an organic molecule has three parts:

- **Numbering** (if necessary)
- **Base**: the parent chain (length and C/C bonding)
 - o First realise that even and odd lengths
- **Suffix**: substituents dictated by the highest priority functional group

2. Start off by identifying the highest priority functional group: this helps you determine the suffix of the molecule's name and can help you determine where to start numbering the parent chain. In subsequent paragraphs, however, I discuss how to begin to identify the parts of the molecule that will be added to the prefix.

There only have C and H atoms left you have hydrocarbons that are based on primary functional groups. The prefix is the key.

3. There begins to find the longest continuous carbon chain you can (the only thing is: branching is good)

- Make sure that you count the longest continuous chain (highest priority functional group) that begins (not ends) at the first carbon
 - o Carbonyls, nitriles, aldehydes, ketones, carboxylic acids
- Find the other side of the chain that most carbon (starting the numbering the next to the longest chain)
- Always include the functional groups and multiple bonds in the structure

4. Begin to number the carbons. The lowest number that you can get that is closest to the side of the parent chain (the longest chain) is the best. The following is a list of the functional groups that are not priority:

- The highest priority functional group
- Any multiple bonds
- The substituents (other than)
- The number of the carbon chain (if necessary)

5. Once you have numbered the parent chain, you can now the appropriate name to indicate the length of the chain.

6. Identify the other functional groups that are present in the parent chain and add them to the name. Additionally, identify other substituents that are (the name of the name)

- Only C/C single bonds are

Naming Molecules Study Guide Answers

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