



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
(Show the ions involved in the compound)



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation has a charge, write the charge in parentheses after the name.
4. If the anion has a charge, write the charge in parentheses after the name.
5. If the compound is a binary compound, write the name of the element first.
6. If the compound is a polyatomic compound, write the name of the polyatomic ion first.

Binary Compounds

First Element	Anion
Na	sodium
Cl	chlorine
O	oxygen
S	sulfur
N	nitrogen
C	carbon
H	hydrogen
F	fluorine
I	iodine
Br	bromine

Common Ions and Their Names

Formula	Name
Na ⁺	sodium
Cl ⁻	chloride
O ²⁻	oxide
S ²⁻	sulfide
N ³⁻	nitride
C ⁴⁻	carbide
H ⁺	hydrogen
F ⁻	fluoride
I ⁻	iodide
Br ⁻	bromide

Naming Compounds with a Metal and a Nonmetal



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.

Naming Anions

Element	Ion Name	Element	Ion Name
Na	Na ⁺	Cl	Cl ⁻
Mg	Mg ²⁺	S	S ²⁻
Ca	Ca ²⁺	O	O ²⁻
Al	Al ³⁺	N	N ³⁻
K	K ⁺	C	C ⁴⁻
Li	Li ⁺	H	H ⁺
Fe	Fe ²⁺	F	F ⁻
Co	Co ²⁺	I	I ⁻
Ni	Ni ²⁺	Br	Br ⁻

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	table salt	NaOH	caustic soda
HCl	hydrochloric acid	Na ₂ CO ₃	soda ash
H ₂ SO ₄	sulfuric acid	NaHCO ₃	baking soda
Na ₂ O	sodium oxide	Na ₂ SO ₄	washing soda

Naming Compounds Study Guide

**Marcia L. Gillette, Ira Blei, Wendy
Gloffke, George Odian**



Naming Compounds Study Guide:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement *The Practice of Chemistry Study Guide & Solutions Manual* Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at approaching new

situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer

Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

Organic Chemistry, 5e Student Study Guide and Solutions Manual David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With *Organic Chemistry Student Study Guide and Solutions Manual* 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's *Basic Concepts of Chemistry* provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter

Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy, 2011

Study Guide and Solutions Manual Neil E. Schore, 2002-08-02

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