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MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Which of the following statements about the atom $^{12}_6\text{C}$ is FALSE? 1) _____
- A) It has 12 neutrons in its nucleus.
 - B) It has 6 protons in its nucleus.
 - C) Its atomic weight is 12.
 - D) It has 6 electrons orbiting the nucleus.
 - E) Its atomic number is 6.

Table 2.1

O C H

- 2) Using the information in Table 2.1, calculate the molecular weight of ethanol, $\text{C}_2\text{H}_5\text{OH}$. 2) _____
- A) 33
 - B) 96
 - C) 46
 - D) 34
 - E) The answer cannot be determined.
- 3) Antacids neutralize acid by the following reaction. Identify the salt in the following equation: 3) _____
- $$\text{Mg}(\text{OH})_2 + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$$
- A) H_2O
 - B) MgCl_2
 - C) HCl
 - D) $\text{Mg}(\text{OH})_2$
 - E) None of the answers is correct.
- 4) Which of the following statements is FALSE? 4) _____
- A) Water freezes from the top down.
 - B) Salts readily dissolve in water.
 - C) Water molecules are formed by hydrolysis.
 - D) Water is a part of a dehydration synthesis reaction.
 - E) Water is a polar molecule.
- 5) Which of the following is the type of bond holding K^+ and I^- ions in KI? 5) _____
- A) hydrogen bond
 - B) ionic bond
 - C) covalent bond
- 6) Which of the following is the type of bond between molecules of water in a beaker of water? 6) _____
- A) hydrogen bond
 - B) covalent bond
 - C) ionic bond

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Arshad Iqbal



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Table of Contents Microbiology Practice Test Questions Tortora 11th Edition

1. Understanding the eBook Microbiology Practice Test Questions Tortora 11th Edition
 - The Rise of Digital Reading Microbiology Practice Test Questions Tortora 11th Edition
 - Advantages of eBooks Over Traditional Books
2. Identifying Microbiology Practice Test Questions Tortora 11th Edition
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Microbiology Practice Test Questions Tortora 11th Edition
 - User-Friendly Interface
4. Exploring eBook Recommendations from Microbiology Practice Test Questions Tortora 11th Edition
 - Personalized Recommendations
 - Microbiology Practice Test Questions Tortora 11th Edition User Reviews and Ratings
 - Microbiology Practice Test Questions Tortora 11th Edition and Bestseller Lists

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 - Interactive Elements Microbiology Practice Test Questions Tortora 11th Edition
8. Staying Engaged with Microbiology Practice Test Questions Tortora 11th Edition
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Microbiology Practice Test Questions Tortora 11th Edition
9. Balancing eBooks and Physical Books Microbiology Practice Test Questions Tortora 11th Edition
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Microbiology Practice Test Questions Tortora 11th Edition
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Microbiology Practice Test Questions Tortora 11th Edition
 - Setting Reading Goals Microbiology Practice Test Questions Tortora 11th Edition
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Microbiology Practice Test Questions Tortora 11th Edition
 - Fact-Checking eBook Content of Microbiology Practice Test Questions Tortora 11th Edition
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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