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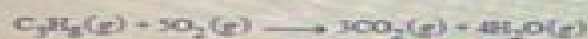
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Your work is overdue! Late penalties now apply (50% day) and you have 1 day left to complete the assignment.

**i** This is a Numeric Entry question. / It is worth 1 point. / You have unlimited attempts. / There is no attempt penalty.

**06 Question** (1 point)[See page 236](#)

Using the balanced equation shown below, calculate the mass of  $C_3H_8$  that must react in order to release  $1.350 \times 10^2$  kJ of heat.



$$\Delta H_{\text{rxn}} = -2219.9 \text{ kJ/mol}$$

4th attempt

[See Periodic Table](#)[See Hint](#)  $\times 10^{\text{}}$  g

&gt; 3rd attempt

&gt; 2nd attempt

&gt; 1st attempt

3 of 20 QUESTIONS COMPLETED

&lt; 06/20 &gt;

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# Norton Smartwork Chemistry Homework Answers

**Tobias Bleicker**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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