

ACTIVITY 2

Name: _____ Class: S6/1 Date: _____ Score: _____

Definitions:

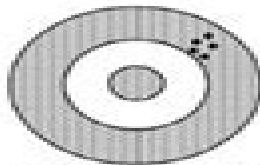
accuracy – how close a measurement is to _____

precision – how close a measurement is to _____

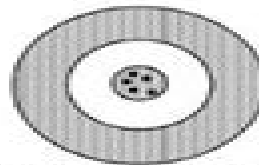
Precision versus Accuracy:

Look at each target below and decide whether the situation is accurate, precise, both, or neither:

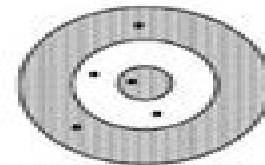
(Note: it is "accepted" that the bull's eye is the place everyone aims for.)



Accurate?: Yes / No
Precise?: Yes / No



Accurate?: Yes / No
Precise?: Yes / No



Accurate?: Yes / No
Precise?: Yes / No

_____ measures Precision

_____ measures Accuracy

Error range = Average \pm range of values

Precision Problems:

Several lab groups measure the density of aluminum. Here is their data:

Team 1	Team 2	Team 3	Team 4	Team 5	Team 6	Team 7
2.65 g/cm ³	2.75 g/cm ³	2.80 g/cm ³	2.77 g/cm ³	2.60 g/cm ³	2.65 g/cm ³	2.68 g/cm ³

What is the average density? _____ g/cm³

Subtract the highest value from the lowest value: _____

Divide this number by 2: _____

The precision of the measurement can be shown as _____ \pm _____ g/cm³.

This is called _____

Observatory Precision And Accuracy Worksheet Answers

SA Dillow



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