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MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question.

- 1) Which of the following statements about the atom $^{12}_6\text{C}$ is FALSE? 1) _____
- A) It has 12 neutrons in its nucleus.
 - B) It has 6 protons in its nucleus.
 - C) Its atomic weight is 12.
 - D) It has 6 electrons orbiting the nucleus.
 - E) Its atomic number is 6.

Table 2.1

O C H

- 2) Using the information in Table 2.1, calculate the molecular weight of ethanol, $\text{C}_2\text{H}_5\text{OH}$. 2) _____
- A) 33
 - B) 96
 - C) 46
 - D) 34
 - E) The answer cannot be determined.
- 3) Antacids neutralize acid by the following reaction. Identify the salt in the following equation: 3) _____
- $$\text{Mg}(\text{OH})_2 + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$$
- A) H_2O
 - B) MgCl_2
 - C) HCl
 - D) $\text{Mg}(\text{OH})_2$
 - E) None of the answers is correct.
- 4) Which of the following statements is FALSE? 4) _____
- A) Water freezes from the top down.
 - B) Salts readily dissolve in water.
 - C) Water molecules are formed by hydrolysis.
 - D) Water is a part of a dehydration synthesis reaction.
 - E) Water is a polar molecule.
- 5) Which of the following is the type of bond holding K^+ and I^- ions in KI? 5) _____
- A) hydrogen bond
 - B) ionic bond
 - C) covalent bond
- 6) Which of the following is the type of bond between molecules of water in a beaker of water? 6) _____
- A) hydrogen bond
 - B) covalent bond
 - C) ionic bond

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H Kauffman



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