

#### CONTENTS OF THIS PACKET

This packet contains the following information regarding mechanic testing, certification, and trainee permit:

- Information for the Mechanic Trainee
- Locating Secretary of State Branch Offices & Test Tips.
- Mechanic Certification Frequently Asked Questions.
- Mechanic Study Guides

Read the information in this packet carefully. Then, if you have further questions about the materials, contact the Licensing Unit at 1-888-SOS-MICH (1-888-767-6424).

#### INFORMATION FOR THE MECHANIC TRAINER

There are two ways a person can get the skills needed to be a mechanic. The first is by going to a school which offers automotive training. This usually includes hands-on learning along with classroom studies. The other way is by working at a repair shop with an experienced mechanic who can leach proper repair methods and see that the work done by the trainee is correct.

Most good mechanics have learned their trade by both schooling and experience on the job. Today's cars and trucks are becoming more and more complicated to repair. This means that both training and experience are more important than ever to the person who wants to earn a living as a mechanic.

If you are currently performing major repairs on motor vehicles in Michigan, you must be a certified mechanic or hold a valid trainee permit.

A mechanic trainee permit is valid for a period of two years from the date of issue in the major repair categories listed on the permit. A permit may <u>not</u> be renewed. Upon expiration of the trainee permit, a mechanic trainee must either become certified or stop performing repairs in the categories listed on the permit.

A mechanic trainee may perform repairs only in the categories listed on the permit under the supervision of a mechanic who is currently certified by the State of Michigan in those repair categories.

# **Michigan Mechanics Certification Study Guide**

**R Pring** 

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1s orbital and the third is placed in the 2s orbital giving an electron configuration of 1s 2s 2s 1

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