





CHARGE YOUR HEADSET

Charge your headset's battery for 2 hours:



 Plug the Motorola charger into the end of the headset.

The indicator light turns red when the battery is charging. It may take up to 1 minute for the indicator light to turn on. When the headset's battery is fully charged (about 2 hours), the indicator light turns green.

Disconnect the charger.

Note: Headset is not functional while charging.



ENSURE YOUR HEADSET IS IN PAIRING MODE

Before you can use your headset, you must pair (link) it with your phone.

Open the headset boom to initiate easy pairing. After a few moments, the indicator light will be steadily lit in purple to indicate headset is in pairing mode.

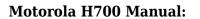
Indicator Light (steadily purple – for pairing mode)



If the headset does not enter pairing mode, see **TROUBLESHOOTING** for help.

Motorola H700 Manual

Bianca Mateo



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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web the pressure volume temperature pvt fluids lab in ku s chemical petroleum engineering department is used to study the properties of fluids under a wide range of pressure and temperature

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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