



## motorola Bluetooth® Handsfree Headset



H700

# MOTOSTART

Get connected in 4 quick steps.

## BEFORE YOU BEGIN

Before using this product, read the Important Safety and Legal Information pamphlet and follow its instructions.

Take a moment before you get started to familiarize yourself with your new H700 Headset.



# 1

## CHARGE YOUR HEADSET

Charge your headset's battery for 2 hours:



- 1 Plug the Motorola charger into the end of the headset.

The indicator light turns red when the battery is charging. It may take up to 1 minute for the indicator light to turn on. When the headset's battery is fully charged (about 2 hours), the indicator light turns green.

- 2 Disconnect the charger.

*Note: Headset is not functional while charging.*

# 2

## ENSURE YOUR HEADSET IS IN PAIRING MODE

Before you can use your headset, you must pair (link) it with your phone.

Open the headset boom to initiate easy pairing. After a few moments, the indicator light will be steadily lit in purple to indicate headset is in pairing mode.

Indicator Light  
(steadily purple  
for pairing mode)



If the headset does not enter pairing mode, see **TROUBLESHOOTING** for help.

# Motorola H700 Manual

**Bianca Mateo**



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1 267 per degree in 1874 victor regnault obtained better experimental results showing this number to be 1 273 and concluded that the pressure would approach zero at 1 273 degrees c

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web figure pageindex 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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web volume pressure kpa linear pressure kpa volume ml pressure kpa in part b we observed the relationship between temperature and pressure the pressure of the 125 ml flask was at 101 36 kpa when the temperature was at 23 2 c

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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