

# Chemistry Reference Tables

Name	Value
Avogadro's number	$6.022 \times 10^{23}$ particles/mole
Gas constant ( $R$ )	$0.0821 \frac{\text{L atm}}{\text{mole K}}$ $62.4 \frac{\text{L mmHg}}{\text{mole K}}$ $8.314 \frac{\text{L kPa}}{\text{mole K}}$
Standard pressure	1.00 atm = 101.3 kPa = 760. mmHg = 760. torr
Standard temperature	0°C or 273K
Volume of 1 mole of any gas at STP	22.4 L

Thermodynamic Constants	Symbol	Value
Heat of fusion of water	$H_f$ (water)	334 J/g
Heat of vaporization of water	$H_v$ (water)	2,260 J/g
Specific heat of water	$C_p$ (water)	$2.05 \frac{\text{J}}{\text{g}^\circ\text{C}}$ for ice, $2.02 \frac{\text{J}}{\text{g}^\circ\text{C}}$ for steam, $4.18 \frac{\text{J}}{\text{g}^\circ\text{C}}$ for liquid

Metal	Specific Heat $\frac{\text{J}}{\text{g}^\circ\text{C}}$	Density (g/cm <sup>3</sup> )	Melting Point (°C)
Aluminum	0.897	2.702	660
Copper	0.385	8.92	1083
Gold	0.129	19.31	1064
Iron	0.449	7.86	1535
Lead	0.129	11.3437	328
Magnesium	1.023	1.74	649
Mercury	0.140	13.5939	-39
Nickel	0.444	8.90	1455
Titanium	0.523	4.5	1660
Zinc	0.388	7.14	420

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