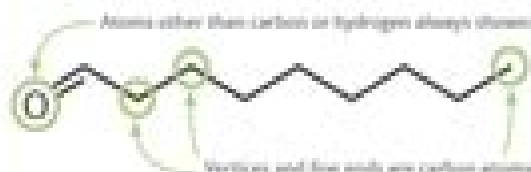


A BASIC GUIDE TO DECODING ORGANIC COMPOUND NAMES

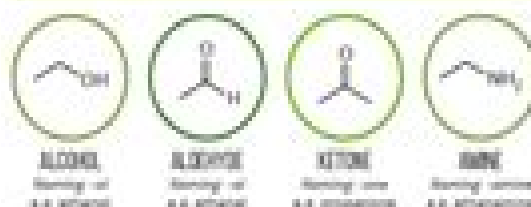
The names of organic molecules can be long and look like a confusing mix of words and numbers. However, they follow a particular set of rules which allows their structure to be decoded from their name. This graphic summarises some basic rules, and shows how they apply to some organic molecules.

ORGANIC COMPOUND REPRESENTATION



Organic molecules are usually represented using skeletal formulae. In these diagrams, the line ends and vertices represent carbon atoms. Hydrogen atoms are 'implied' – that is, they are not usually shown, but each carbon must have four bonds, and it's assumed they have the required number of hydrogens for this to be the case. Atoms other than carbon or hydrogen are always shown, and hydrogen atoms are shown if they are bonded to one of these 'heteroatoms'.

FUNCTIONAL GROUPS



A molecule's functional group is the group of atoms that give it its chemical properties and reactivity. It's usually indicated by a suffix at the end of the name, with a number indicating its position if this is required for clarity. There are many different functional groups.

Different functional groups have different suffixes. Alcohols (-ol), aldehydes (-al), & ketones (-one) are examples of functional groups.

BOND TYPES



Carbon atoms can be linked by single bonds, double bonds, or even triple bonds. The name of the molecule reflects the bonds present.

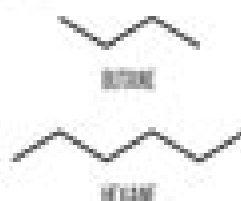
- an- present in name = molecule contains only single bonds
- en- present in name = molecule contains at least 1 double bond
- yn- present in name = molecule contains at least 1 triple bond

For double and triple bonds, numbers indicate their position.

PARENT CHAIN

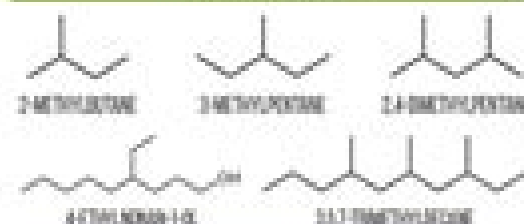
NUMBER OF CARBONS DENOTED BY PREFIX

- | | |
|---------|---------|
| 1 METH- | 6 HEX- |
| 2 ETH- | 7 HEPT- |
| 3 PROP- | 8 OCT- |
| 4 BUT- | 9 NON- |
| 5 PENT- | 10 DEC- |



Part of the organic molecule's name denotes how many carbons make up its 'parent chain'. This is defined as the longest continuously connected chain of carbon atoms including the functional group in the molecule. Carbons not included are dealt with as 'side chains'.

SIDE CHAINS



Molecules can have one or more carbons that aren't part of the parent chain, referred to as 'side chains'. The number of carbons in the side chain is used to name it, in the same way as for the parent chain, but the ending -yl is then added. A number is added to show the location of the side chain on the parent chain. If there is more than one of the same side chain at different points, the prefixes di- (2), tri- (3), or tetra- (4) are used in the name.

STEREISOMERISM

2 MAIN TYPES: E-Z ISOMERISM (DOUBLE) D-OPTICAL ISOMERISM (CHIRAL)



Chemical names sometimes contain a letter in brackets, for example (2S), (3R), (4R), or (5S). These refer to stereoisomerism when a molecule has the same chemical formula as another, but a different arrangement in 3D space. This can be due to a different arrangement of atoms around a double bond, or when a molecule has two different arrangements of four different groups of atoms around a central carbon which are non-superimposable mirror images.



Organic Compounds Notetaking Guide

Susan Wise Bauer, Jessie Wise



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Patterson, 1927 Student Guide for Workplace Monitor Training: Basic industrial hygiene, 1985 *The Composting Handbook* Robert Rynk, Ginny Black, Jane Gilbert, Johannes Biala, Jean Bonhot, Mary Schwarz, Leslie Cooperband, 2021-12-03 The Composting Handbook provides a single guide to the science principles and best practices of composting for large scale composting operations facing a variety of opportunities and challenges converting raw organic materials into a useful and marketable product Composting is a well established and increasingly important method to recycle and add value to organic by products Many if not most of the materials composting treats are discarded materials that would otherwise place a burden on communities industries farms and the environment Composting converts these materials into a valuable material compost that regenerates soils improving soils for plant growth and environmental conservation The Composting Handbook expands on previously available resources by incorporating new information new subjects and new practices drawing its content from current scientific principles research engineering and industry experience In both depth and breadth it covers the knowledge that a compost producer needs to succeed Topics include the composting process methods of composting equipment site requirements environmental issues and impacts business knowledge safety and the qualities uses and markets for the compost products The Composting Handbook is an invaluable reference for composting facility managers and operators prospective managers and operators regulators policy makers environmental advocates educators waste generators and managers and generally people interested in composting as a business or a solution It is also appropriate as a textbook for college courses and a supplemental text for training courses about composting or organic waste management Created in conjunction with the Compost Research and Education Foundation CREF Includes the latest information on composting and

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