FINAL EXAM Summary Sheet

Potential Study Questions

<u>Disclaimer</u>: The following are potential topics for exam questions and sample questions; the exam may or may not contain these questions, and questions not included in this sheet may or may not be on the exam. There also may be typos in this sheet.

ALKANES

Physical Properties

- Hydrocarbons, dispersion forces, non-polar, not miscible w/water, low density
- sp³ hybridized carbon: C-C, C-H bonds, electronegativity, MO theory & orbitals
- m.p.: odd/even effect.
- b.p. C1-4: gases, C5-17: liquids, C18 and up: solids
- linear, branched, cyclics, chirality (incl. optical activity, D/L, R/S, d/l,+/-),
- non-equivalent carbons
- isomers: constitutional, conformational isomers, stereo-isomers
- naming, special groups:
 - o Me, Et, Pr, Bu
 - o iPr, iBu, sec-Bu, tBu
 - o neopentyl

<u>Task</u>: Name compounds 1 and 2; identify any chiral centers and # of chemically non-equivalent carbons; where would the compounds absorb in IR in UVvis? Identify the participating orbitals.



- Rotational/conformational isomers = conformations
 - o eclipsed
 - values: H/H, CH3/H, CH3/CH3
 - o staggered, anti, gauche
 - value: gauche: CH3/CH3

Task: draw an energy rotational energy plot (360°) for pentane along C2-C3 bond

- Cycloalkanes:
 - o Spiro, fused, bridged: naming
 - o Heats of combustion: ring strains (angle, torsional, sterics)
 - o Cis/trans
 - 1,3-diaxial interactions
 - value H/CH3
 - Ring-flip for cyclohexane

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Neil E. Schore

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Table of Contents Organic Chem Final Study Guide

- 1. Understanding the eBook Organic Chem Final Study Guide
 - The Rise of Digital Reading Organic Chem Final Study Guide
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Organic Chem Final Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Organic Chem Final Study Guide
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Organic Chem Final Study Guide
 - Personalized Recommendations

- Organic Chem Final Study Guide User Reviews and Ratings
- o Organic Chem Final Study Guide and Bestseller Lists
- 5. Accessing Organic Chem Final Study Guide Free and Paid eBooks
 - Organic Chem Final Study Guide Public Domain eBooks
 - o Organic Chem Final Study Guide eBook Subscription Services
 - o Organic Chem Final Study Guide Budget-Friendly Options
- 6. Navigating Organic Chem Final Study Guide eBook Formats
 - o ePub, PDF, MOBI, and More
 - o Organic Chem Final Study Guide Compatibility with Devices
 - Organic Chem Final Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of Organic Chem Final Study Guide
 - Highlighting and Note-Taking Organic Chem Final Study Guide
 - o Interactive Elements Organic Chem Final Study Guide
- 8. Staying Engaged with Organic Chem Final Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Organic Chem Final Study Guide
- 9. Balancing eBooks and Physical Books Organic Chem Final Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Organic Chem Final Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chem Final Study Guide
 - Setting Reading Goals Organic Chem Final Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chem Final Study Guide
 - Fact-Checking eBook Content of Organic Chem Final Study Guide

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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