

C. Melting Point Determination

Recrystallized Benzoic acid	Trials	Melting Point (°C)	Complete Liquidation (°C)	Melting point range (°C)	Change in color	Shrinking of size
	1	122	127	122-127	Simultaneously white to colorless	Gradually shrinks up to 100°C up to melting point
	2	122	127	122-127	Simultaneously white to colorless	Gradually shrinks up to 100°C up to melting point

Benzoic samples	Melting point range (°C)
Recrystallized	122-127
Crude	126-128

- As the crystals were heated, there is a continuous but gradual shrinking of the size of the crystals as evidenced by the diminishing or decreasing height of the sample observed in the capillary tubes for both trials accompanied by a change in color as indicated. Also, it was observed that after removal of the capillary tubes after each trial from the heat oil bath, the purified Benzoic crystals immediately solidified, that is, returned to the original solid phase.

IV. Discussion and Interpretation of Results

Recrystallization from a solvent depends primarily on the fact that more solids are more soluble in hot solvents than cold solvents, and second, different materials are soluble in different solvents in various solvents. In order for recrystallization to be successful, there should be a large difference in the solubility of the solid material in hot solvents compared to cold solvents. A material can be purified by recrystallization when both the desired substance and the impurity have similar solubilities, but only when the impurity represents a small fraction of the total solid. The desired substance will crystallize on cooling, but the impurity will not.

The most crucial and time-consuming feature of recrystallization is the selection of solvent because important considerations should be observed. In order to choose a good recrystallization solvent, the solubility of the solute should be compared at its other words, sparingly soluble only in the solvent at room temperature. Thus, from this relative description, we have to consider the principle "like dissolves like" which refers to the polarity of both the solute and the solvent. Substances of same polarity tend to dissolve one another more at room temperature, which therefore contributes the very first condition of choosing a good recrystallization solvent.

In the experiment, benzoic acid, whose structural formula is given below, was used as the sample.

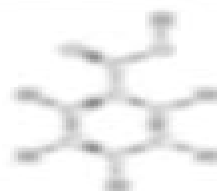


Figure 1. Structural formula of Benzoic acid

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Michal Rosen-Zvi



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Cooperative Chemistry Lab Manual Cooper, 2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations

Experimental Organic Chemistry Joaquín Isac-García, José A. Dobado, Francisco G. Calvo-Flores, Henar Martínez-García, 2015-10-30 Experimental Organic Chemistry Laboratory Manual is designed as a primer to initiate students in Organic Chemistry laboratory work Organic Chemistry is an eminently experimental science that is based on a well established theoretical framework where the basic aspects are well established but at the same time are under constant development Therefore it is essential for future professionals to develop a strong background in the laboratory as soon as possible forming good habits from the outset and developing the necessary skills to address the challenges of the experimental work This book is divided into three parts In the first safety issues in laboratories are addressed offering tips for keeping laboratory notebooks In the second the material the main basic laboratory procedures preparation of samples for different spectroscopic techniques Microscale Green Chemistry and qualitative organic analysis are described The third part consists of a collection of 84 experiments divided into 5 modules and arranged according to complexity The last two chapters are devoted to the practices at Microscale Synthesis and Green Chemistry seeking alternatives to traditional Organic Chemistry Organizes lab course coverage in a logical and useful way Features a valuable chapter on Green Chemistry Experiments Includes 84 experiments arranged according to increasing complexity Microscale Inorganic Chemistry Zvi Szafran, Ronald M. Pike, Mono M. Singh, 1991-01-24 A comprehensive treatment of the subject of microscale inorganic chemistry is provided through 45 laboratory experiments These include experiments in main group and transition metal chemistry instrumental techniques kinetics synthesis and the manipulation of air sensitive material Practical Manual Organic & Medicinal Chemistry A Rama Narsimha Reddy, L Srividya, The purpose of this Manual of Practical Organic and Medicinal Chemistry is intended for the Pharmacy students of D Pharm B Pharm and Pharm D students as per the new regulations of PCI 2016 This is specifically written to meet the present needs of revised curriculum **Laboratory Manual for Fundamentals of General, Organic, and Biological Chemistry** John R. Holum, Ruth C. Denison, 1978 **Applied Chemistry | AICTE Prescribed Textbook - English** Anju Rawlley, D. V. Saraf, 2021-11-01 This text book o Applied Chemistry is development as per AICTE model curriculum 2018 for compulsory course on Applied Chemistry of first years Diploma Programme in Engineering and Technology Atomic Structure Chemical Bonding Solution Water Engineering

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