



2. In solvent like H_2O



Vicinal dibromide formation :

Halohydrin formation

- It is a **regioselective** reaction
- The nucleophilic attack of the water occurs at the more substituted carbon atom.

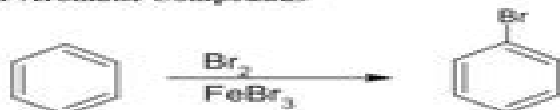
2. With Alkynes

- It is an **anti-addition** reaction:
- 1. In solvent like CCl_4 , CH_2Cl_2 etc



Tetrabromide formation

3. With Aromatic Compounds

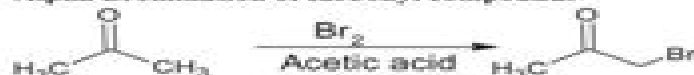


4. Photo halogenation of alkanes



Major

5. Alpha Bromination of carbonyl compounds



6. Oxidation of Monosaccharides

1. Bromine water is a mild oxidizing agent
2. selectively oxidizes the aldehyde group (for example CHO group of glucose) to a carboxylic acid.



7. Haloform Reaction

1. converts methyl ketones to Haloform (CHX_3 ($X=Cl, Br, I$)) and salt of carboxylic acids.

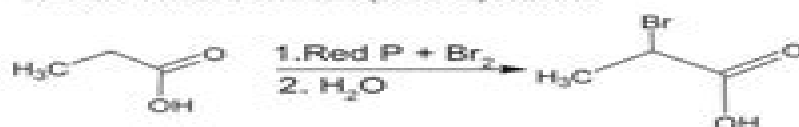


8. The Hoffmann Bromamide degradation

1. It converts primary amide to a primary amine with one less carbon atom.



9. Hell-Volhard-Zelinski (or HVZ) reaction



5. CN^-

Cyanide Ion and Nitrile Group

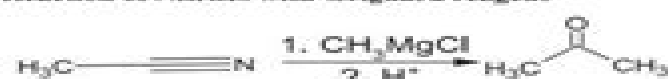
1. Conversion of Nitriles to a Primary Amine



2. Conversion of Nitriles to Aldehydes



3. Reaction of Nitriles with Grignard reagent



4. Hydrolysis of Nitriles to Carboxylic Acids



Organic Chemistry Mechanism Study Guide

Roman Wölfel



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central theme of the authors approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us

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