

Name \_\_\_\_\_ Date \_\_\_\_\_ Class \_\_\_\_\_

**CHAPTER 15** **STUDY GUIDE FOR CONTENT MASTERY**

## Solutions

### Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air's (4) soluble substance. A substance that does not dissolve is (5) insoluble. (6) Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

**Solvation is the process of surrounding solute particles with solvent particles to form a solution.**

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**CHAPTER 15** **STUDY GUIDE FOR CONTENT MASTERY**

### Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

**The sodium and chloride ions are separated and surrounded by the water molecules.**

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

**Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.**

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

**The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.**

13. List three ways that the rate of solvation may be increased.

**stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent**

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

**true** 14. The overall energy change that occurs when a solution forms is called the heat of solution.

**false** 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

**false** 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

**true** 17. In a saturated solution, solvation and crystallization are in equilibrium.

**true** 18. Additional solute can be dissolved in an unsaturated solution.

**true** 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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# **Mole Study Guide For Content Mastery Answer**

**Adrianne Dill Linton, Mary Ann  
Matteson**



## **Mole Study Guide For Content Mastery Answer:**

**Study Guide and Laboratory Manual for Physical Examination and Health Assessment, Canadian Edition- E-Book** Carolyn Jarvis, Ann L. Eckhardt, 2023-02-23 NEW Preface includes an overview of the NCSBN Clinical Judgement Measurement Model PN competencies and the REx PN NEW Clinical judgement exercises equip students for success on the Next Generation NCLEX NGN including questions with an increased focus on clinical judgement Updated content throughout corresponds to the 4th Canadian edition of the textbook and incorporates the latest research and evidence informed practice

, **Introduction to Maternity & Pediatric Nursing - E-Book** Gloria Leifer, 2013-11-28 Part of the popular LPN Threads series Introduction to Maternity Pediatric Nursing provides a solid foundation in obstetrics and pediatric nursing An easy to follow organization by developmental stages discussion of disorders by body system from simple to complex and health to illness and a focus on family health make it a complete guide to caring for maternity and pediatric patients Written in a clear concise style by Gloria Leifer MA RN this edition reflects the current NCLEX test plan with additional material on safety health promotion nutrition and related psychosocial care Cultural Considerations boxes and a Cultural Assessment Data Collection Tool help in developing individualized plans of care Updated health promotion content includes Health Promotion boxes focusing on preventive strategies for achieving prenatal wellness health during pregnancy postnatal health and pediatric illness prevention and wellness including the complete immunization schedules for all ages Nursing Tips provide information applying to the clinical setting Objectives are listed in each chapter opener Key terms include phonetic pronunciations and text page references at the beginning of each chapter Nursing Care Plans with critical thinking questions help you understand how a care plan is developed how to evaluate care of a patient and how to apply critical thinking skills A companion Evolve website includes animations videos answers to review questions and answer guidelines for critical thinking questions an English Spanish audio glossary critical thinking case studies and additional review questions for the NCLEX examination

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**Foundations of College Chemistry** Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a

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Biology Sylvia S. Mader,2000-07      **The Advocate** ,2004-08-17 The Advocate is a lesbian gay bisexual transgender LGBT monthly newsmagazine Established in 1967 it is the oldest continuing LGBT publication in the United States

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