

Chemistry 12

Solutions Manual Part B

Unit 3 Energy Changes and Rates of Reaction

Solutions to Practice Problems in Chapter 5 Energy Changes

Calculating the Absorption of Heat (Student textbook page 281)

1. How much heat must be added to a 36.2 g sample of iron to increase its temperature by 250.0°C?

What Is Required?

You have to calculate the amount of heat that must be added to a sample of iron to increase its temperature by 250.0°C.

What Is Given?

You know the mass of iron: $m = 36.2 \text{ g}$

You know the change in the temperature: $\Delta T = 250.0^\circ\text{C}$

You have the specific heat capacity of iron from **Table 5.1** on page 280 of the student text: $c_{\text{Fe}} = 0.449 \text{ J/g}^\circ\text{C}$

Plan Your Strategy	Act on Your Strategy
Substitute the given information into the formula $Q = mc\Delta T$. Solve for Q .	$\begin{aligned} Q &= mc\Delta T \\ &= (36.2 \text{ g}) \left(0.449 \frac{\text{J}}{\text{g}^\circ\text{C}} \right) (250.0^\circ\text{C}) \\ &= 4063.45 \text{ J} \\ &= 4.06 \text{ kJ} \end{aligned}$ The iron must gain 4060 J or 4.06 kJ of heat energy.

Check Your Solution

Check that the information has been substituted correctly. The answer has the correct number of significant digits.

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A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the text 'New Zealand. Mines Department'.

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Nuclear Science Abstracts, 1974 **Electro-Chemo-Mechanics of Solids** Sean R. Bishop, Nicola H. Perry, Dario Marrocchelli, Brian W. Sheldon, 2017-03-18 This book brings together a collection of chapters that focus on the relationship among electrical chemical and mechanical properties and the study of adjusting one property through the control of another namely Electro Chemo Mechanics ECM The authors examine how this relationship can result in beneficial properties such as mixed ionic and electronic conductivity in oxides upon oxygen deficiency or lithium insertion electro chemo and or changes in ionic and electronic mobility observed in strained systems electro mechano They also consider how ECM interactions can be responsible for large stresses from non stoichiometry induced lattice dilation chemo mechano While many volumes are available devoted to the study of the origins and characteristics of electro chemical relationships they form the well known field of electrochemistry this volume is highly novel in its examination of the corresponding electro mechanical chemo mechanical and electro chemo mechanical relationships The book is ideal for researchers and design engineers interested in energy storage and conversion and the electrical and mechanical properties of materials **Energy Research Abstracts**

,1992-12 **Nitric Oxide Research from Chemistry to Biology: EPR Spectroscopy of Nitrosylated Compounds** Yann A. Henry, Annie Guissani, Beatrice Ducastel, 2012-12-06 *Abstract Bulletin of the Institute of Paper Chemistry* Institute of Paper Chemistry (Appleton, Wis.), 1961 Ultrasound Technologies for Food and Bioprocessing Hao Feng, Gustavo Barbosa-Canovas, Jochen Weiss, 2010-11-17 Traditional food and bioprocessing technologies are facing challenges due to high expectation from the consumers and producers for better quality and safety higher process efficiency and products with novel properties or functionalities For this reason in the last few years new forms of physical energies have been explored to propose alternatives to traditional processing technologies Acoustic energy has the potential to replace or partially substitute conventional processes and at the same time offer unique opportunities in the characterization of foods and biomaterials This book is a resource for experts and newcomers in the field of power ultrasound gives insights into the physical principles of this technology details the latest advancements and links them to current and potential applications in the food and bioprocessing related industries **Journal of the Chemical Society** Chemical Society (Great Britain), 1965 Titles of chemical papers in British and foreign journals included in Quarterly journal v 1 12 **Mines Statement** New Zealand. Mines Department, 1906 **Papers and Reports Relating to Minerals and Mining** New Zealand. Mines Department, 1906 **Papers and Reports Relating to Minerals and Mining** New Zealand. Mines Dept, 1906 Appendix to the Journal of the House of the Representatives New Zealand. Legislature. House of Representatives, 1906 Appendix to the Journals of the House of Representatives of New Zealand New Zealand. Parliament. House of Representatives, 1906 **First National Symposium on Nuclear Chemistry, Warsaw, Feb. 10-12, 1964** ,1965 **ERDA Energy Research Abstracts** ,1983 Journal of Organic Chemistry of the USSR ,1976

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