

A Contemporary Guide to the Therapeutic Use of Essential Oils



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100 Science-related terms and definitions

1. **Atom** - The smallest particle of an element which cannot be created or destroyed.
2. **Molecule** - A group of atoms joined together by chemical bonds.
3. **Compound** - A substance formed by the chemical combination of two or more elements in a fixed ratio.
4. **Mixture** - A substance formed by the physical combination of two or more substances in any ratio.
5. **Element** - A substance which cannot be broken down into simpler substances by chemical means.
6. **Chemical change** - A change in which a new substance is formed with different properties from the original substances.
7. **Physical change** - A change in which no new substance is formed and the original substance can be recovered back.
8. **Chemical reaction** - A process in which one or more substances are converted into one or more new substances.
9. **Chemical equation** - A symbolic representation of a chemical reaction.
10. **Reactants** - The substances that take part in a chemical reaction.
11. **Products** - The substances formed as a result of a chemical reaction.
12. **Chemical formula** - A symbolic representation of the composition of a compound.
13. **Valency** - The combining capacity of an element.
14. **Atomic number** - The number of protons in the nucleus of an atom.
15. **Mass number** - The sum of protons and neutrons in the nucleus of an atom.
16. **Isotopes** - Atoms of the same element having the same atomic number but different mass numbers.
17. **Electron** - A negatively charged subatomic particle.
18. **Proton** - A positively charged subatomic particle.
19. **Neutron** - A neutral subatomic particle.
20. **Atomic structure** - The arrangement of protons, neutrons, and electrons in an atom.
21. **Electron shell** - A region around the nucleus where electrons are present.
22. **Atomic size** - The distance from the nucleus to the outermost shell of an atom.
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