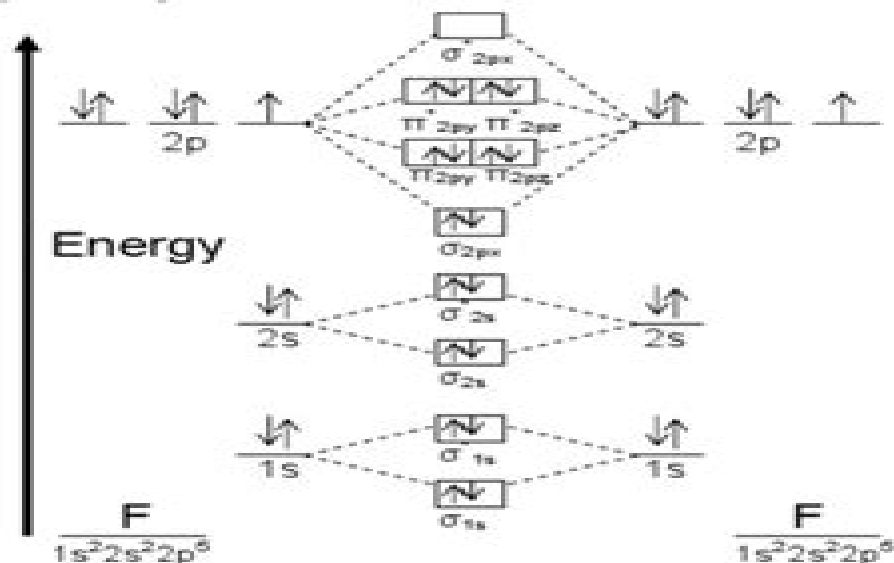


These problems are for practice in drawing your molecular orbital diagrams, molecular electron configurations and determining bond order.

The following questions pertain to the  $F_2$  molecule:

- A) Draw the molecular orbital energy diagram for this molecule. Label all of the orbitals specifically.



- B) Give the molecular electron configuration for the molecule:



- C) Determine the bond order for the molecule:

$$\text{Bond Order} = (10 - 8) / 2 = 1$$

- D) Indicate whether the species is paramagnetic or diamagnetic:

Diamagnetic: No unpaired electrons

- F) Compare the relative stability of this molecule to  $F_2^+$  and  $F_2^-$ :



# Molecular Orbital Practice Answers

**J Elliott**



## **Molecular Orbital Practice Answers:**

## Unveiling the Magic of Words: A Report on "**Molecular Orbital Practice Answers**"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Molecular Orbital Practice Answers**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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