

1. Moles and chemical equations :

- Law of conservation of mass :

Total mass of reactants is equal to the total mass of products.

(equation must be balanced)

*Note :

When you write the balanced equation, the coefficients of balancing are mole ratios between reactants and products.

Examples :



0.5 mole of Al, How many moles Al_2O_3 ?

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Whitten



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