

## Extension Questions

15. One mole of any gas will occupy 22.4 L of volume at standard temperature and pressure (STP).

Consider this situation as it relates to the reaction in Model 1:  $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$

a. Calculate the volumes of reactants consumed and the volume of product made.

b. Record the ratio of  $\text{N}_2$  to  $\text{H}_2$  to  $\text{NH}_3$ . Reduce the ratio to the lowest whole numbers possible.

	$\text{N}_2$ Consumed	$\text{H}_2$ Consumed	$\text{NH}_3$ Produced	Volume Ratio $\text{N}_2:\text{H}_2:\text{NH}_3$
How many liters of each substance would be consumed or produced if the reaction occurred $6.02 \times 10^{23}$ times at STP?	22.4 L	67.2 L	44.8 L	1 : 3 : 2

16. Refer to the data table in Question 15.

a. Can the mole ratio from a balanced chemical equation be interpreted as a ratio of volumes for gases?

Yes. In this case, the mole ratio matches with volume.

b. Use mathematical concepts to explain how your answer in part a is possible.

We are multiplying by the same number (22.4 L), so we can divide by 22.4 to get back to the ratio of 1 : 3 : 2.

17. Explain why the ratio of volumes is NOT followed in the following reactions.



44.8 L      22.4 L      0.036 L



22.4 L      22.4 L      0.035 L

The volume ratio for volumes of a gas and these two reactions produces a liquid and a solid.

18. Which of the following quantities are **conserved** (total amount in reactants = total amount in products) in a chemical reaction? Find an example or counter-example from this activity to support your answer for each.

a. Molecules - **does not conserve**

$$1 : 3 : 2 \quad 1 + 3 \neq 2$$

c. Mass - **conserves**

$$19 + 5 = 19 \quad 19 + 5 = 19$$

e. Atoms of an element - **conserves**



$$2\text{N} + 6\text{H} = 2\text{N} + 6\text{H}$$

b. Moles - **does not conserve**

$$1 : 3 : 2 \quad 1 + 3 \neq 2$$

d. Volume - **does not conserve**

$$1 : 1 : 2 \quad 1 + 1 \neq 2$$

$$(22.4 + 22.4 \neq 44.8)$$

# Mole Ratio Pogil Answer Key

**NARAYAN CHANGDER**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

### **Mole Ratio Pogil Answer Key:**

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## **Table of Contents Mole Ratio Pogil Answer Key**

1. Understanding the eBook Mole Ratio Pogil Answer Key
  - The Rise of Digital Reading Mole Ratio Pogil Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Mole Ratio Pogil Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Mole Ratio Pogil Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Mole Ratio Pogil Answer Key
  - Personalized Recommendations
  - Mole Ratio Pogil Answer Key User Reviews and Ratings
  - Mole Ratio Pogil Answer Key and Bestseller Lists
5. Accessing Mole Ratio Pogil Answer Key Free and Paid eBooks
  - Mole Ratio Pogil Answer Key Public Domain eBooks
  - Mole Ratio Pogil Answer Key eBook Subscription Services

- Mole Ratio Pogil Answer Key Budget-Friendly Options
- 6. Navigating Mole Ratio Pogil Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Mole Ratio Pogil Answer Key Compatibility with Devices
  - Mole Ratio Pogil Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Mole Ratio Pogil Answer Key
  - Highlighting and Note-Taking Mole Ratio Pogil Answer Key
  - Interactive Elements Mole Ratio Pogil Answer Key
- 8. Staying Engaged with Mole Ratio Pogil Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Mole Ratio Pogil Answer Key
- 9. Balancing eBooks and Physical Books Mole Ratio Pogil Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Mole Ratio Pogil Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Mole Ratio Pogil Answer Key
  - Setting Reading Goals Mole Ratio Pogil Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mole Ratio Pogil Answer Key
  - Fact-Checking eBook Content of Mole Ratio Pogil Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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