

Administrator's Handbook

Motorola\* Embedded Software Version 9.0.6

# Motorola NVG510 Voice Gateway







# **Motorola Nvg 510 Manual**

**R Barnett** 

## **Motorola Nvg 510 Manual:**

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web isolation of trimyristin from nutmeg purpose the purpose of this experiment is to isolate myristic acid from ground nutmeg seeds nutmeg seeds are extracted using thme to yield crude trimyristin then recrystallized to yield pure trimyristin which will be hydrolyzed and undergo esterification

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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyristin is a triglyceride what are triglycerides prelab question 2 how do microwaves heat microwave procedure add approximately 2 5 g of ground nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web in this week s lab experiment the purpose of this lab is to new techniques isolate thechemical trimyristin from the common spice nutmeg this lab requires five techniques simple distillation vacuum filtration melting point determination reflux and extraction

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

 $isolation\ of\ trimyristin\ from\ nutmeg\ name\ report\ studocu\ -\ Jan\ 11\ 2023$ 

web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

lab 4 extraction of trimyristin from nutmeg university of toronto - Sep 07 2022

web in this lab trimyristin is extracted from ground nutmeg occuring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

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