

Learning Objectives

Chapter 1: Energy Levels, Transitions, Spectra

Block 1: 100 – 104 questions on following:

- 1) Hydrogen-like atoms: comparing the spectrum of hydrogen with that of hydrogen-like spectra of other elements. What do they tell about the atomic structure and spectra?
- 2) Effect of spin-orbit coupling on energy levels and transitions. How does it affect the spectrum? (Consider the effect of spin-orbit coupling on the spectrum of hydrogen-like atoms.)
- 3) Effect of the Pauli exclusion principle on the spectrum of hydrogen-like atoms. How does it affect the spectrum?
- 4) Spectroscopic selection rules: transitions between energy levels. How does it affect the spectrum?

Block 2: 105 – 109 questions on following:

- 1) Hydrogen-like atoms: comparing the spectrum of hydrogen with that of hydrogen-like spectra of other elements. What do they tell about the atomic structure and spectra?
- 2) Effect of spin-orbit coupling on energy levels and transitions. How does it affect the spectrum? (Consider the effect of spin-orbit coupling on the spectrum of hydrogen-like atoms.)
- 3) Effect of the Pauli exclusion principle on the spectrum of hydrogen-like atoms. How does it affect the spectrum?
- 4) Spectroscopic selection rules: transitions between energy levels. How does it affect the spectrum?

Block 3: 110 – 114 questions on following:

- 1) Hydrogen-like atoms: comparing the spectrum of hydrogen with that of hydrogen-like spectra of other elements. What do they tell about the atomic structure and spectra?
- 2) Effect of spin-orbit coupling on energy levels and transitions. How does it affect the spectrum? (Consider the effect of spin-orbit coupling on the spectrum of hydrogen-like atoms.)
- 3) Effect of the Pauli exclusion principle on the spectrum of hydrogen-like atoms. How does it affect the spectrum?
- 4) Spectroscopic selection rules: transitions between energy levels. How does it affect the spectrum?

Block 4: 115 – 119 questions on following:

- 1) Hydrogen-like atoms: comparing the spectrum of hydrogen with that of hydrogen-like spectra of other elements. What do they tell about the atomic structure and spectra?
- 2) Effect of spin-orbit coupling on energy levels and transitions. How does it affect the spectrum? (Consider the effect of spin-orbit coupling on the spectrum of hydrogen-like atoms.)

Motivation And Emotion Psychology Study Guide Answers

Richard O. Straub



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