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the quantity of heat for a process is represented by the letter q the sign of q for an endothermic process is positive because the system is gaining heat a chemical reaction or physical

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the quantities commonly encountered in chemical problems include the number of moles of a substance the number of atoms molecules or formula units of a substance and the mass in

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it is always a good idea to estimate the answer before you do the actual calculation in this case the mass given 35 00 g is less than the molar mass so the answer should be less than 1

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the atomic mass of an element expressed in grams is the mass of a mole of the element the mass of a mole of an element is the molar mass to calculate the molar mass of a compound

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web you might need periodic table the first two ionization energies for beryllium are shown below be g be g e be g be 2 g e i
1 900 kJ mol i 2 1757 kJ mol which of the following identifies the most probable value for the third ionization energy for be
and provides the best justification

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