17: Molecular Geometry - VSEPR

Key Structure Terms

- Valence Shellt Electrons in the outermost shell that are involved in bonding.
- Lewis Structure: A 2D representation of a molecule and it's boards.
- . Lone Pair: Pair of electrons not being shared in a bond.
- . Bonding Pairs Pair of electrons used to make a bond. Both atoms sharing the electrons can "count" them in their WINDOWS CO. NOW-CO.
- · Single bond: One pair of shared electrons.
- . Double bond: Two pairs of shared electrons. Shorter and stronger than a single bond.
- . Triple bond: Three pairs of shared electrons. Shorter and stronger than a double bond.
- Valence Shell Electron Pair Repulsion Theory (VSEPR): Bonds and lone pairs are electrons, all electrons are negative, negative things repel other negative things, therefore all bonds and lone pairs arrange themselves in 3D as far away from each other as possible.
- · Electron Geometry: 3D structure of a molecule determined by counting the electron regions around a central atom (bonds and lone pairs).
- · Electron Region: Each bond (single, double or triple) and lone pair count as "1" electron region.
- . Molecular Geometry: 3D structure determined by the atoms bonded to the central atom.
- . Linand: Atoms bonded to the central atom.

VSEPR Theory and Geometry

Bectron geometry is determined by looking at the number of electron regions around the central atom.

Molecular geometry is determined by looking at the number of atoms bonded to the central atom (ligand) and the number of lone pairs around the central atom.

Bectronic Geometry Mnemonic: Linear(2) - Trigonal Planar(3) - Tetrahedral (4) - Irigonal Bipyramidal(5) -Octahedron(6) = "Long InP To InBe Overseas."

Examples:

Electron geometry: Tetrahedron Molecular geometry: Tetrahedron

Electron geometry: Tetrahedron Molecular geometry: Trigonal pyramidal

Molecular Geometries

Electron Regions	Molecular Formula	Name	Shape
3	AX;	Limear	
	(BeCl ₂ , CD ₂)		
3	(BF ₃ , CO ₃), NO ₃ , SO ₃)	Trigonal Planar	
	AX aE	Bent	
	(NO ₂ *, SO ₂ , O ₂)		-
4	(CH., NH.*, PO.*, SO.*,	Tetrahedron	
	(NH _b , H ₂ O*, PCl _b , SO ₂ *)	Trigonal pyramidal	1
	AX ₂ E ₂	Dent	
	(H ₂ O, ClO ₂), OF ₂ , ScO ₂)		
5	AX ₃ (PCI ₅)	Trigonal bipyvamidal	
	(SF., SCL)	See-saw	
	AX, E, (CIF ₂ , ICI ₃)	T-shaped	-
	AX ₃ E ₃ (XeF ₃ , I ₃)	Linear	
6	AX ₄ (SF ₆ , PCl ₆)	Octahedron	
	AX JE (BrFs, IFs)	Square pyramidal	
	AX.E2 (XeF.)	Square planer	- 2

How to Use This Cheat Sheet: These are the keys related to this topic. Try to read through it carefully twice then write it out on a blank sheet of paper. Review it again before the exams.

Molecular Structure And Bonding Cheat Sheet

D. Sangeeta, John R. LaGraff

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common polymerization techniques Polycondensation and non chain polymerizations including dendrimer synthesis and various click procedures are covered in Volume 5 Volume 6 focuses on several aspects of controlled macromolecular architectures and soft nano objects including hybrids and bioconjugates Many of the achievements would have not been possible without new characterization techniques like AFM that allowed direct imaging of single molecules and nano objects with a precision available only recently An entirely new aspect in polymer science is based on the combination of bottom up methods such as polymer synthesis and molecularly programmed self assembly with top down structuring such as lithography and surface templating as presented in Volume 7 It encompasses polymer and nanoparticle assembly in bulk and under confined conditions or influenced by an external field including thin films inorganic organic hybrids or nanofibers Volume 8 expands these concepts focusing on applications in advanced technologies e.g. in electronic industry and centers on combination with top down approach and functional properties like conductivity Another type of functionality that is of rapidly increasing importance in polymer science is introduced in volume 9 It deals with various aspects of polymers in biology and medicine including the response of living cells and tissue to the contact with biofunctional particles and surfaces The last volume is devoted to the scope and potential provided by environmentally benign and green polymers as well as energy related polymers They discuss new technologies needed for a sustainable economy in our world of limited resources Provides broad and in depth coverage of all aspects of polymer science from synthesis polymerization properties and characterization methods and techniques to nanostructures sustainability and energy and biomedical uses of polymers Provides a definitive source for those entering or researching in this area by integrating the multidisciplinary aspects of the science into one unique up to date reference work Electronic version has complete cross referencing and multi media components Volume editors are world experts in their field including a Nobel Prize winner **Polymer Modification** John Meister, 2000-07-25 Describes new modification methods and applications for natural synthetic thermoplastic and thermoset polymers that result from economic forces commercial processes and the latest research and development Features chemical and physical technologies such as sulfonation alkylation acid base hydrolysis hydrogenation stress orienting anneal

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