

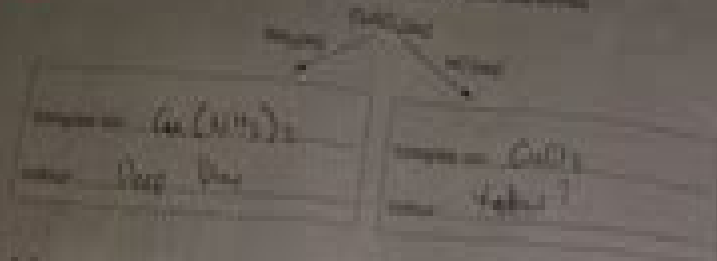
Question 1

Answer all the questions

The reaction is about the synthesis of copper complexes and ligands.

(a) The reaction shows the synthesis of a copper complex.

In the boxes, write the formula and name of the ligands and metal.



(b)  $Cu^{2+}$  has two d-orbitals in  $d_{sp^2}$  hybridization state and two water molecules. The  $d_{sp^2}$  hybridization is a tetrahedral ligand.

The resulting formula of the complex is shown in Fig. 1.1 below.



Fig. 1.1

(c) What is meant by the term tetrahedral ligand?

A ligand that forms four bonds with central metal ion.

(d) The reaction is shown in Fig. 1.2 below.

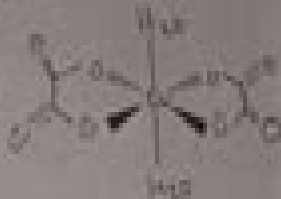

The reaction is about the synthesis of a copper complex. The charge has been given.



Complex A

Complete the 3D diagram of the other two geometries of A. You do not need to include any charges.

Indicate with ticks whether the stereoisomers are cis, trans, optical or a combination of these types.

<p><b>Stereoisomer</b></p>														
<p><b>Type</b></p>	<table border="1" data-bbox="1244 1023 1404 1170"> <tr> <td>cis</td> <td></td> </tr> <tr> <td>trans</td> <td>✓</td> </tr> <tr> <td>optical</td> <td></td> </tr> </table>	cis		trans	✓	optical		<table border="1" data-bbox="1584 1023 1744 1170"> <tr> <td>cis</td> <td>✓</td> </tr> <tr> <td>trans</td> <td></td> </tr> <tr> <td>optical</td> <td>✓</td> </tr> </table>	cis	✓	trans		optical	✓
cis														
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(e) What is the empirical formula, including the charge, of the complex in A?

$[Cu(NH_3)_2(H_2O)_2]^{2+}$

# Ocr Chemistry A2 Unofficial Mark Scheme 2014

**Frank Harriss**



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