

Chemistry C4-Chemical Economics

- 1. The pH scale and chemical modification:



pH	Chemical	
Increasingly acidic	0-1	strongest acids
	1-2	strong acids
	2-3	strong acids
	3-4	strong acids
	4-5	weak acid / strong acids
	5-6	weakest acids
	7	neutral
Increasingly alkaline	8-9	strong bases
	9-10	strong bases
	10-11	strong bases
	11-12	strong bases
	12-13	strong bases
	13-14	strongest bases

- 2. The modification is just a step that changes colour:
 - a) This step in modification changes colour depending on the pH
 - b) Chemical modification is a very useful combination of dye which gives the colour, then the colour change
 - c) It is useful for monitoring the pH of a solution
- 3. The pH is a measure of the acidity of a solution (pH 0-14)
 - a) The pH is a measure of the acidity of a solution (pH 0-14)
 - b) The pH is a measure of the acidity of a solution (pH 0-14)
 - c) The pH is a measure of the acidity of a solution (pH 0-14)
- 4. The pH is a measure of the acidity of a solution (pH 0-14)
 - a) The pH is a measure of the acidity of a solution (pH 0-14)
 - b) The pH is a measure of the acidity of a solution (pH 0-14)
 - c) The pH is a measure of the acidity of a solution (pH 0-14)
- 5. The pH is a measure of the acidity of a solution (pH 0-14)
 - a) The pH is a measure of the acidity of a solution (pH 0-14)
 - b) The pH is a measure of the acidity of a solution (pH 0-14)
 - c) The pH is a measure of the acidity of a solution (pH 0-14)

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Clemens Wendtner

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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