

# Oxidation Numbers

Summary: An atom connected to a **more** eN atom = +1

An atom connected to a **less** eN atom = -1

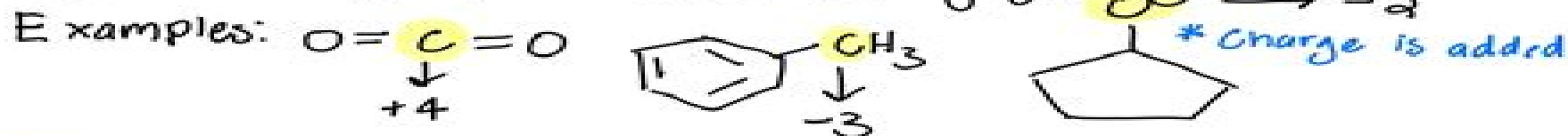
An atom connected the **same** atom = 0

\* Number of bonds must be taken into account

\* Charge is also incorporated into oxidation number

Oxidation number  $\uparrow$  = Oxidized = reducing agent

Oxidation number  $\downarrow$  = reduced = oxidizing agent



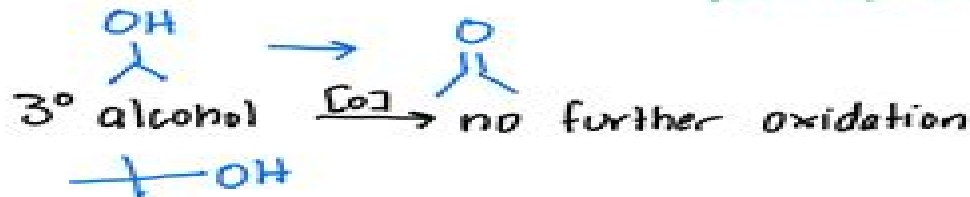
## Oxidation Reactions

Summary: 1° alcohol  $\xrightarrow{[\text{O}]}$  aldehyde  $\xrightarrow[\text{H}_2\text{O}]{\text{must be wet } [\text{O}]}$  carboxylic acid

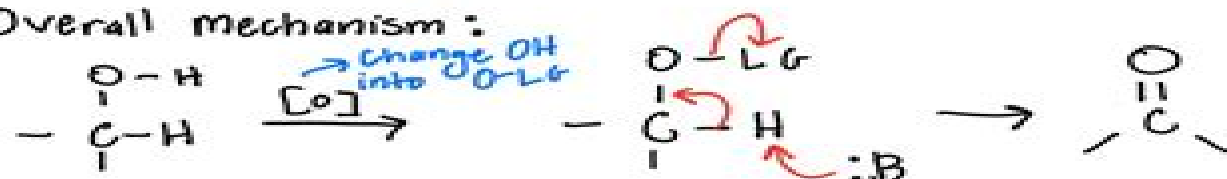


1° alcohol  $\xrightarrow{[\text{O}]}$  aldehyde must be dry

2° alcohol  $\xrightarrow{[\text{O}]}$  ketone can be either wet or dry



Overall mechanism:



# Organic Chem Phenol Study Guide

**Brent L. Iverson**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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