Organic Chemistry 2 Lab - Final Exam Study Guide

- What kind of rxn is the Williamson Ether rxn?
- SN2
 What species served as the nucleophile in the Williamson Ether lab?
 - 4-phenyl phenol
- 3. What are the dangers of benzyl chloride?
- acute toxicity, corrosive, combustible
- 4. Where would the benzyl proton signals be in H-NMR?
 - 6.0-8.0 ppm
- 5. What are the most common leaving groups in the Williamson ether synthesis?
 - bromide, chloride, and tosylate
- 6. What components are used in the Williamson ether synthesis to create an ether?
 - deprotonated alcohol and an electrophile that is an organic molecule with a good leaving group
- 7. What is the reactivity order of electrophile in SN1 rxns?
 - tertiary, secondary, primary (worst)
- 8. What is the order of reactivity for electrophilics in SN2 reactions?
- primary, secondary, tertiary
 What kind of solvent is used in the production of SN2?
 - polar aprotic (DMSO, acetone)
- 10. What kind of solvent is utilized in the production of SN1?
 - polar protic (alcohols)
- 11. What is the difference between nucleophiles for SN1 and SN2?
 - SN1 requires a weak nucleophile and SN2 requires a strong nucleophile that is usually bearing a negative charge
- 12. What are some examples of common oxidizing agents?
 - PCC, H2CrO4, DMP, Swern
- 13. What is the primary MTBE safety concern?
- flammable
- T/F: The bleach targets the oxygen of isoborneol in the chemical process for its oxidation.
 - False; Because isoborneol's oxygen is more electronegative than bleach's chlorine, it attacks the bleach.
- 15. What happens during the sublimation process?
 - a solid turns into a gas
- 16. If the oxidation of isoborneol is completed, what wavelength will there be a peak in the IR?

Organic Chemistry Ii Lab Guide

Dr. Priyanka Gupta Manglik

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