

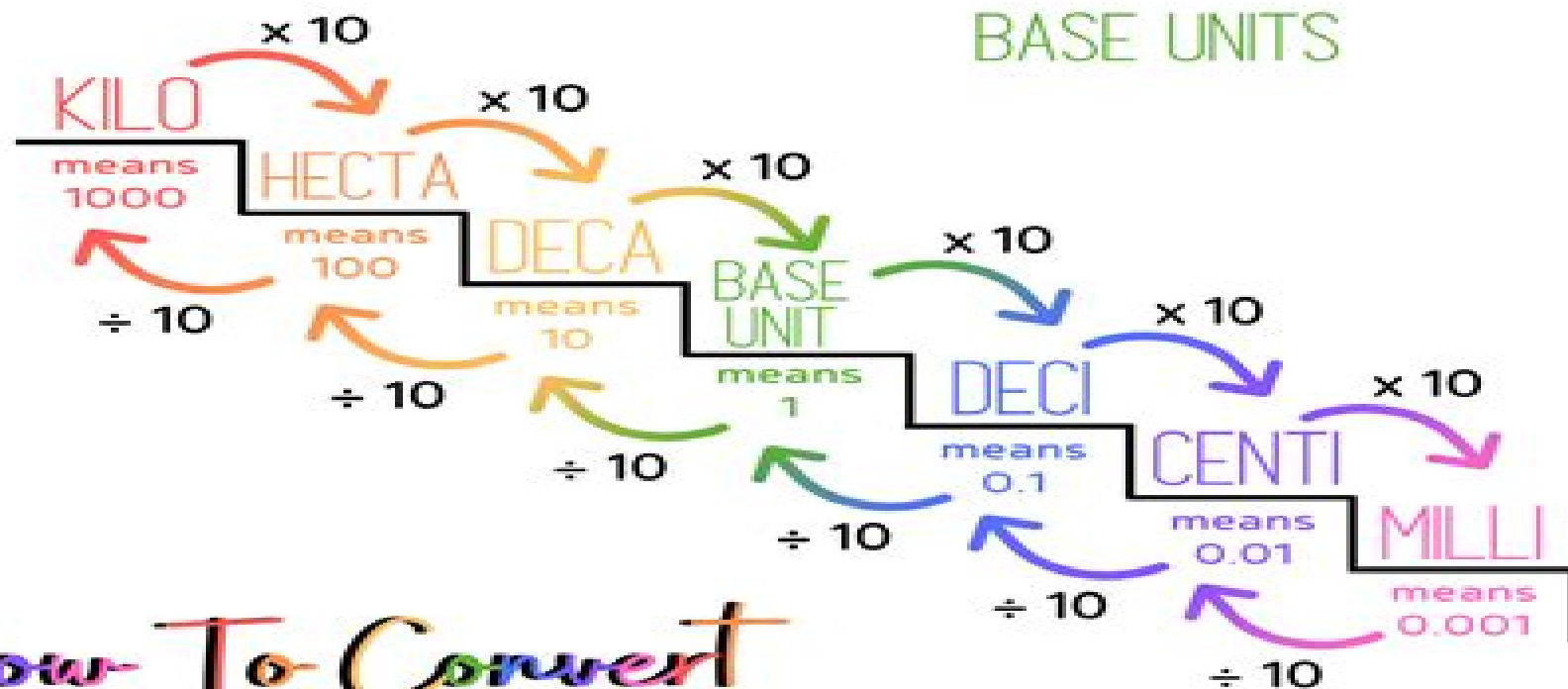
# Metric Units

LENGTH  
meter

WEIGHT  
gram

VOLUME  
liter

BASE UNITS



## How To Convert

move the decimal point in the original number the direction and number of spaces shown in the ladder

### EXAMPLE

convert 0.56 kilometers to meters

meters is 3 steps right of kilometers, so move the decimal point 3 places right

$$0.560 \text{ km} = 560 \text{ m}$$

The diagram shows the number 0.560 with the decimal point moved three places to the right to become 560. A curved arrow indicates the movement from the original decimal point to its new position after the conversion.

# Metric Conversion Study Guide

**Michael Seilmaier**



## **Metric Conversion Study Guide:**

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

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*U.S. Metric Study Report* United States. National Bureau of Standards,1971

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