

- \_\_\_\_ 7. A kilogram(s) contains how many grams?
- $0.1 \times 10^3$
  - $0.1 \times 10^6$
  - 0.1
  - 0.001
  - $0.1 \times 10^{-3}$
- \_\_\_\_ 8. Convert 0.0002 m to mm.
- 200 mm
  - $2 \times 10^{-5} \text{ mm}$
  - $2 \times 10^{-4} \text{ mm}$
  - $2 \times 10^{-3} \text{ mm}$
  - 2 mm
- \_\_\_\_ 9. A conversion factor that shows the relationship between grams and kilograms is
- $\frac{1000 \text{ g}}{1 \text{ kg}}$
  - $\frac{1 \text{ kg}}{1000 \text{ g}}$
  - $\frac{1000}{1}$
  - $\frac{1}{1000}$
  - $\frac{1 \text{ g}}{1000 \text{ kg}}$
- \_\_\_\_ 10. We generally report a measurement by recording all of the certain digits plus \_\_\_\_\_ uncertain digit(s).
- 100
  - 1000
  - 10000
  - 100000
  - 1000000
- \_\_\_\_ 11. In the measurement 0.0031, which digit is the estimated digit?
- 0
  - the 0 immediately to the left of the 3
  - 3
  - the 0 further left of the decimal point
- \_\_\_\_ 12. How many significant figures are there in the number 0.00456789?
- 4
  - 5
  - 7
  - 8
  - 9

# New Mexico End Of Course Chemistry Exam

**Albert A Gayle**



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