

**Table K
Selected Polyatomic Ions**

Formula	Name	Formula	Name
ClO_4^-	perchlorate	ClO_2^-	chlorite
ClO_3^-	chlorate	ClO_2^+	dichlorine oxide
NH_4^+	ammonium	MnO_4^-	permanganate
$\text{C}_2\text{H}_3\text{O}_2^-$ CH_3COO^-	acetate	NO_2^-	nitrite
CN^-	cyanide	NO_3^-	nitrate
CO_3^{2-}	carbonate	O_2^{2-}	peroxide
HCO_3^-	bicarbonate hydrogencarbonate	OH^-	hydroxide
$\text{C}_2\text{O}_4^{2-}$	oxalate	PO_4^{3-}	phosphate
F^-	fluoride	SCN^-	thiocyanate
Br^-	bromide	SO_3^{2-}	sulfite
I^-	iodide	SO_3^{2-}	sulfite
ClO_2^-	chlorite	HSO_4^-	hydrogensulfate
ClO_3^-	chlorate	$\text{S}_2\text{O}_3^{2-}$	thiosulfate
ClO_4^-	perchlorate		

**Table L
Solubility Guidelines for Aqueous Solutions**

Most That Form Soluble Compounds	Exceptions	Most That Form Insoluble Compounds*	Exceptions
Group 1 ions (Li^+ , Na^+ , etc.)		acetate (CH_3CO_2^-)	when combined with Group 1 ions or ammonium (NH_4^+)
ammonium (NH_4^+)		chromate (CrO_4^{2-})	when combined with Group 1 ions, Ca^{2+} , Sr^{2+} , or ammonium (NH_4^+)
nitrate (NO_3^-)		phosphate (PO_4^{3-})	when combined with Group 1 ions or ammonium (NH_4^+)
halides (Cl^- , Br^- , or I^-)		oxide (O^{2-})	when combined with Group 1 ions or ammonium (NH_4^+)
hydrogencarbonate (HCO_3^-)		sulfide (S^{2-})	when combined with Group 1 ions or ammonium (NH_4^+)
acetate (CH_3CO_2^-)		hydroxide (OH^-)	when combined with Group 1 ions, Ca^{2+} , Sr^{2+} , or ammonium (NH_4^+)
halides (Cl^- , Br^- , I^-)	when combined with Ag^+ , Pb^{2+} , or Hg_2^{2+}		
sulfate (SO_4^{2-})	when combined with Ag^+ , Ca^{2+} , Sr^{2+} , Ba^{2+} , or Pb^{2+}		

*Anions with Group 1 ions are soluble in H_2O .

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