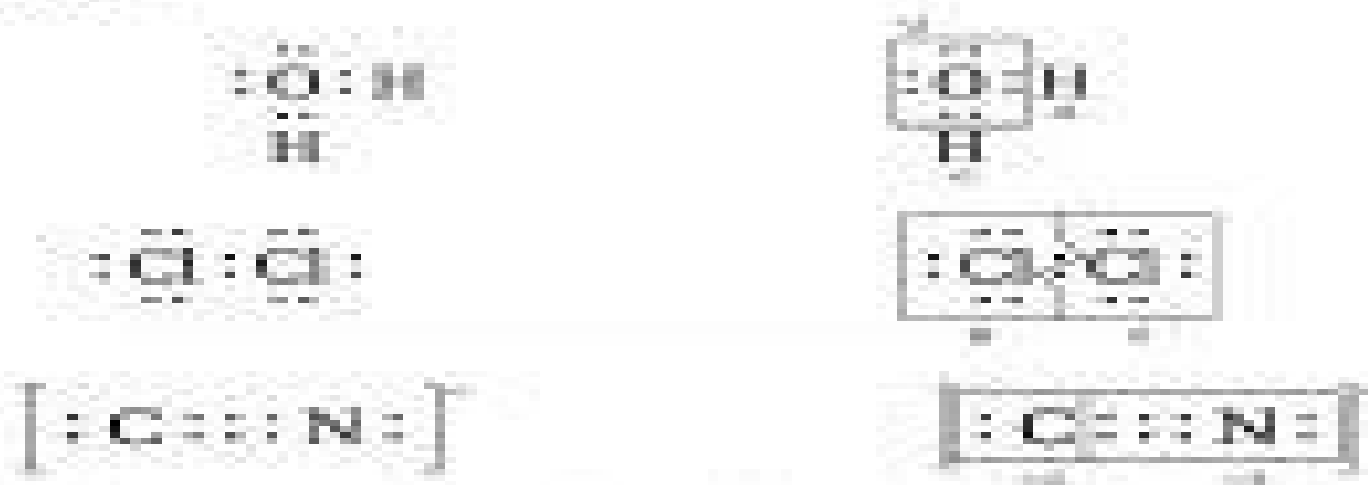


# Resonance-Structural Drawing

## Resonance Structures:



1. correct electron-dot notation for single or multiple covalent bonds; each electron is represented by a dot (electron pair) or

electron pairs are represented by a pair of dots;

2. Lewis dot notation;

3. the structure is enclosed in brackets for covalent compounds; and

4. a structural formula is drawn on top of an electron-dot diagram or vice versa.

5. equivalent structures of the ion are separated by a plus sign.

6. or

7. the Lewis structure should be drawn on top of

electron-dot notation; and

8. the structure is enclosed in brackets for ionic species.

9. cations or anions have appropriate charge symbols or ions without charge are indicated without any symbol.

10. or

11. the ion is completely drawn (electron-dot notation) and

12. the electron-dot diagram is

indicated on top within appropriate brackets for the ion.

13. ionization of the ion is indicated by a plus or minus sign.

14. the Lewis structure is drawn on top of the electron-dot diagram.



# Mhr Chemistry 12 Aswers

**Shasha Hu**



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