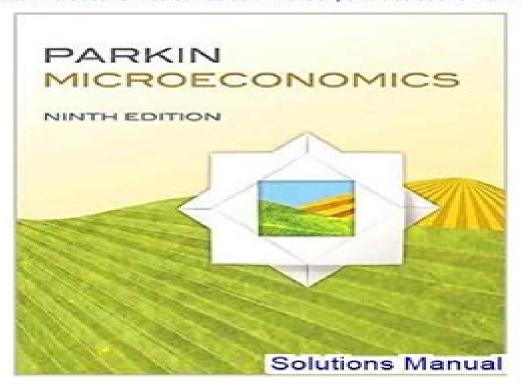
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# **Microeconomics Parkin Solution Manual Chapter 11**

**Robin Bade, Michael Parkin** 

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web understand that parts of an acid base titration be able to determine the k a or k b from ph data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate k a1 and k a2 for a polyprotic acid by the end of this lab students should be able to

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web answer based on the data the strong acid strong base combination of hydrochloric acid and sodium hydroxide had the

neutral ph at the equivalence point being 7 this result is expected because a combination of strong acids and strong bases should result in a neutral at the equivalence point

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