

## WORKSHEET – ASSIGNING OXIDATION NUMBERS

Name \_\_\_\_\_

Period \_\_\_\_\_

### Oxidation Number Rules:

1. A pure element has an oxidation number of 0.
2. The oxidation number of an element in a monatomic ion equals the charge of the ion.
3. The sum of the oxidation number of all the elements in a compound equals 0.
4. The sum of the oxidation numbers of all the elements in a polyatomic ion equals the charge on the ion.

Directions: In the following questions, give the oxidation number of the indicated atoms/ion.



# Oxidation Numbering Problems And Answers

**B Lingard**



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