



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Two Elements
 (Name the ions separately and then combine)

- EXAMPLE**
- Binary Ionic Compounds**
- Directions for naming:**
- Show the cation first.
 - Show the anion second.
 - Use the full name of the cation.
 - Use the full name of the anion.
 - Use the charge of the anion to determine the charge of the cation.
 - Use the charge of the cation to determine the charge of the anion.
 - Use the charge of the cation to determine the charge of the anion.
 - Use the charge of the anion to determine the charge of the cation.

Cations		Anions	
Full Name	Symbol	Full Name	Symbol
Hydrogen	H ⁺	Fluoride	F ⁻
Lithium	Li ⁺	Chloride	Cl ⁻
Sodium	Na ⁺	Bromide	Br ⁻
Potassium	K ⁺	Iodide	I ⁻
Ammonium	NH ₄ ⁺	Sulfide	S ²⁻
Calcium	Ca ²⁺	Selenide	Se ²⁻
Strontium	Sr ²⁺	Telluride	Te ²⁻
Barium	Ba ²⁺	Stannide	Sn ²⁻
Lead	Pb ²⁺	Plumbide	Pb ²⁻
Aluminum	Al ³⁺	Antimonide	Sb ³⁻
Zinc	Zn ²⁺	Bismuthide	Bi ³⁻
Iron	Fe ²⁺	Phosphide	P ³⁻
Copper	Cu ²⁺	Arsenide	As ³⁻
Mercury	Hg ²⁺	Nitride	N ³⁻
Gold	Au ³⁺	Oxide	O ²⁻
Silver	Ag ⁺	Sulfate	SO ₄ ²⁻
Hydrogen	H ⁺	Sulfite	SO ₃ ²⁻
Lithium	Li ⁺	Carbonate	CO ₃ ²⁻
Sodium	Na ⁺	Phosphate	PO ₄ ³⁻
Potassium	K ⁺	Acetate	CH ₃ COO ⁻
Ammonium	NH ₄ ⁺	Perchlorate	ClO ₄ ⁻
Calcium	Ca ²⁺	Chlorate	ClO ₃ ⁻
Strontium	Sr ²⁺	Chlorite	ClO ₂ ⁻
Barium	Ba ²⁺	Hypochlorite	ClO ⁻
Lead	Pb ²⁺	Perbromate	BrO ₄ ⁻
Aluminum	Al ³⁺	Bromate	BrO ₃ ⁻
Zinc	Zn ²⁺	Bromite	BrO ₂ ⁻
Iron	Fe ²⁺	Hypobromite	BrO ⁻
Copper	Cu ²⁺	Periodate	IO ₄ ⁻
Mercury	Hg ²⁺	Iodate	IO ₃ ⁻
Gold	Au ³⁺	Iodite	IO ₂ ⁻
Silver	Ag ⁺	Hypoiodite	IO ⁻

Naming Compounds with a Metal and a Nonmetal

- EXAMPLE**
- Binary Molecular**
- Directions for naming:**
- Write the name of the first element.
 - Write the name of the second element.
 - Use the prefix to show the number of atoms of each element.
 - Use the suffix to show the number of atoms of each element.

Prefixes			
Number	Prefix	Number	Prefix
1	mono	10	deca
2	di	11	undeca
3	tri	12	duodeca
4	tetra	13	tredec
5	penta	14	quattuordec
6	hexa	15	quindec
7	hepta	16	sexdec
8	octa	17	septendec
9	nona	18	octadec

Common Names for Specific Compounds			
Formula	Common Name	Formula	Common Name
CO	Carbon monoxide	CO ₂	Carbon dioxide
NO	Nitric oxide	NO ₂	Nitrogen dioxide
NO ₂	Nitrogen dioxide	N ₂ O	Dinitrogen oxide
N ₂ O	Dinitrogen oxide	N ₂ O ₄	Dinitrogen tetroxide
SO ₂	Sulfur dioxide	SO ₃	Sulfur trioxide
SO ₃	Sulfur trioxide	SO ₂ Cl ₂	Sulfuryl chloride
SO ₂ Cl ₂	Sulfuryl chloride	SO ₂ F ₂	Sulfuryl fluoride
SO ₂ F ₂	Sulfuryl fluoride	Cl ₂ O	Dichlorine monoxide
Cl ₂ O	Dichlorine monoxide	Cl ₂ O ₂	Dichlorine dioxide
Cl ₂ O ₂	Dichlorine dioxide	Cl ₂ O ₃	Dichlorine trioxide
Cl ₂ O ₃	Dichlorine trioxide	Cl ₂ O ₄	Dichlorine tetroxide
Cl ₂ O ₄	Dichlorine tetroxide	Cl ₂ O ₅	Dichlorine pentoxide
Cl ₂ O ₅	Dichlorine pentoxide	Cl ₂ O ₆	Dichlorine hexoxide
Cl ₂ O ₆	Dichlorine hexoxide	Cl ₂ O ₇	Dichlorine heptoxide
Cl ₂ O ₇	Dichlorine heptoxide		

Naming Compounds Study Guide

Kenneth J. Hughes, Paul B. Kelter



Naming Compounds Study Guide:

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

The Practice of Chemistry Study Guide & Solutions Manual Pamela Mills, Amina El-Ashmawy, 2003-04-14 Designed to help students understand the material better and avoid common mistakes Also includes solutions and explanations to odd numbered exercises

Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

Organic Chemistry, Student Study Guide and Solutions Manual David R. Klein, 2017-01-04 This is the Student Study Guide and Solutions Manual to accompany Organic Chemistry 3e Organic Chemistry 3rd Edition is not merely a compilation of principles but rather it is a disciplined method of thought and analysis Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems Readers must learn to become proficient at

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Series,Mark Kernion,Joseph A. Mascetta,2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein,Laurie S. Starkey,2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With *Organic Chemistry Student Study Guide and Solutions Manual 5th Edition* students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry Basic Concepts of Chemistry, Study Guide Leo J. Malone,1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone,Theodore O. Dolter,2012-01-03 The 9th edition of Malone s *Basic Concepts of Chemistry* provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Hazmat Chemistry Study Guide (Second Edition) Jill Meryl Levy,2011

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Study Guide, Introductory Chemistry, a Foundation ; Introductory Chemistry ; Basic Chemistry Iris Stovall, Donald J. DeCoste, 2000

Student Study Guide to Accompany Chemistry Kenneth J. Hughes, Paul B. Kelter, 2002-05 Reviews essential points from the text presents problems and hints to assist the student in solving them Solutions to selected problems in the end of chapter problem sets are included By Kenneth Hughes of the University of Wisconsin Madison

Naming Compounds Study Guide Book Review: Unveiling the Power of Words

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