

# Orion pH meter 420A

## Guía Rápida

### Auto calibración con 2 Buffers

1. Escoja buffers estándar, ya sea 7.00 y 4.01 o 7.00 y 10.01, cualesquiera que se asocien al rango en pH requerido en la muestra en cuestión.
2. Presione la tecla **mode** las veces que sea necesario hasta que el modo de **pH** sea indicado por el puntero en la pantalla (éste es el modo predeterminado).
3. Coloque el electrodo dentro de una solución buffer estándar pH 7.00.
4. Presione **2nd**, luego **cal**. La palabra **CALIBRATE** debería aparecer en la parte superior central de la pantalla. El tiempo y la fecha de la última calibración son mostrados; Los valores reportados usualmente tienen una ligera relación con la realidad. Después de algunos segundos **P1** es mostrado en la parte inferior de la pantalla, indicando que el lector está listo para la primera solución estándar pedirá confirmación [**" P "** for **" Buffer ?"**] (**" P "** a Buffer **"?**), cuándo piense en esto, elija [**" Don't ask "**] (**" No Preguntar "**) porque el electrodo ya debe estar ahí. Cuando el electrodo este leyendo y la lectura sea establezca la máquina emite un **BIP**, el indicador **READY** aparecerá, y el valor de temperatura correcta de la solución buffer es mostrada. No haga nada hasta que usted escuche el bip, o de lo contrario el programa de calibración se interrumpe, nunca, nunca aterrice el electrodo mientras lea. Sea paciente. Si no emite el **BIP**, quiere decir que la lectura del electrodo todavía no se estabiliza, lo cual usualmente indica un problema con el electrodo. Una vez que usted oye el **BIP** y ve en la pantalla **READY**, presione **YES**, para aceptar de este valor corregido con temperatura. La lectura se mostrará por unos instantes, luego **P2** aparecerá en la parte inferior de la pantalla, indicando que el lector está listo para la segunda solución buffer.
5. Enjuague el electrodo con agua del des ionizada, luego colóquelo en la segunda solución buffer, ya sea pH 4.01 o 10.01, dependiendo del rango en el cual se quiera trabajar. Espere el **BIP**. Sea paciente. Una vez que escuche el **BIP** y ve en la pantalla **READY**, Oprima **YES**, para aceptar del en segundo valor corregido con temperatura. Después de que el segundo valor de la solución buffer haya sido aceptado, el aumento en el electrodo será mostrará. **SLP** aparece en la parte inferior de la pantalla mientras el aumento real del electrodo en porcentaje se muestra en el campo principal de la pantalla. Indican valores "aceptables" los que mienten entre 80 % y 120 %; Los valores fuera de este rango son incorrectos, cualquiera que sea el valor. El medidor automáticamente cambia al modo de **MEASURE**. **MEASURE** indica que ahora está listo para tomar lectura de la muestra deseada; la lectura será mostrada arriba en el campo principal.
6. Enjuague el electrodo y colóquelo en la muestra a evaluar. La lectura del pH se aprecia directamente en la pantalla; La temperatura de muestra será indicada en el campo inferior de la pantalla.



# Orion 420a User Manual

**Zelpha B. Johnson,D. Wayne Kellogg**



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Thomas Scientific ,2000      *Evaluation of Methods for Determination of Ammonia in Glass Cleaners and Other Consumer Products* Liesa Caldwell Houdashelt,1995      **East European Accessions Index** ,1958      *Trends in Colloid and Interface Science XIII* Durdica Tezak,Mladen Martinis,2003-07-01 This volume includes a number of selected papers of the 12th Conference of the European Colloid and Interface Society held in September 1998 in Dubrovnik and Cavtat Croatia The topics included are Amphiphiles Monolayers and Micelles Solutions and Suspensions Emulsions and Microemulsions Polymers Interfaces and Experimental techniques      **Mechanical and Aerospace Engineering, ICMAE2011** Wu Fan,2011-10-24 Selected peer reviewed papers from the 2nd International Conference on Mechanical and Aerospace Engineering ICMAE 2011 July 29 31 2011 Bangkok Thailand      **The Official Railway Guide** ,1877      *Remediation and Management of Degraded Lands* MH Wong,2018-05-04 Remediation and Management of Degraded Lands presents the program of the first International Conference on the Remediation and Management of Degraded Lands This collection reviews the extent of resource debasement and offers solutions for their restoration The 14 part first section deals with mine management and rehabilitation Topics include the devastating results of open cut mining open pit mining lignite surface mining and acid mining Despite such ruin the articles reveal the possibilities for reclamation Part two devotes nine chapters to the management of derelict lands Reforestation soil fertility prognosis and the uses of nitrogen are just a few of the covered subjects This portion of the book pays special attention to the successful results of remediation in China and Hong Kong The final division addresses soil contamination and reclamation There are eleven chapters on subjects that include the single and interactive effects of aluminum the effectiveness of EDTA HCI and the value of pig on litter compost as a tool for edible crop growth These and other innovative techniques make Remediation and Management of Degraded Lands a valuable addition to any environmental library      Proceedings of the International Topical Meeting on Nuclear and Hazardous Waste Management, Spectrum '96 ,1996      Vanadium: The Versatile Metal Kenneth Kustin,Joao Costa-Pessoa,Debbie C. Crans,2007 This book describes recent developments in techniques for studying vanadium new vanadium compounds and new advances in fundamental studies of vanadium and includes applications to the enzymology toxicology and anit carcinogenic and insulin enhancing abilities of vanadium compounds      Effect of Cheese on Red Wine Flavor Berenice Madrigal Galan,2005      *Free Radical Research* ,2003      **American Journal of Enology and Viticulture** ,2007      **Journal of the Chemical Society of Pakistan** Chemical Society of Pakistan,2006      **World Gold '97** Australasian Institute of Mining and Metallurgy,1997      *Effects of Canopy Manipulation on Cabernet Sauvignon Composition and Flavor* Jimena Balic,2006      **Arkansas Animal Science Department Report 2002** Zelpha B. Johnson,D. Wayne Kellogg,2002      *Research Series* ,1985      Aqueous Polymer — Cosolute Systems Dan F. Anghel,2003-04-01 This collection of formulas has been written by applied scientists and industrial engineers for design professionals and students who work in

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